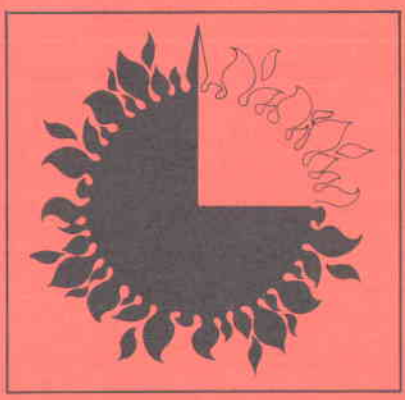


E37

JPL - 5102-105

Thermal Power Systems
Advanced Solar Thermal Technology Project

Glass for Solar Concentrator Applications



April 1, 1979

Prepared for
U.S. Department of Energy
by
Jet Propulsion Laboratory
California Institute of Technology
Pasadena, California

5102-105

Thermal Power Systems
Advanced Solar Thermal Technology Project

Glass for Solar Concentrator Applications

F.L. Bouquet

April 1, 1979

Prepared for
U.S. Department of Energy
by
Jet Propulsion Laboratory
California Institute of Technology
Pasadena, California

Prepared by the Jet Propulsion Laboratory, California Institute of Technology, for the U.S. Department of Energy through an agreement with the National Aeronautics and Space Administration.

The JPL Solar Thermal Power Systems Project is sponsored by the U.S. Department of Energy and forms a part of the Solar Thermal Program to develop low-cost solar thermal electric generating plants.

This report was prepared as an account of work sponsored by the United States Government. Neither the United States nor the United States Department of Energy, nor any of their employees, nor any of their contractors, subcontractors, or their employees, makes any warranty, express or implied, or assumes any legal liability or responsibility for the accuracy, completeness or usefulness of any information, apparatus, product or process disclosed, or represents that its use would not infringe privately owned rights.

PREFACE

This work was performed by the Applied Mechanics
Division of Jet Propulsion Laboratory, California
Institute of Technology.

ACKNOWLEDGMENT

The author gratefully acknowledges the contributions from many individuals from industry and government that made this report possible. Special recognition is given to Sandia Corp., McDonnell Douglas Astronautics-West, and Corning Glass Works. Recognition is given to members of the JPL solar-thermal power development team, including Dr. Marc Adams and Hugh Maxwell of the JPL Materials Development Group.

ABSTRACT

Materials for highly reflective surfaces for application to solar thermal power systems are treated in this report. The primary consideration is on the use of second-surface glass mirrors with comparison to alternate candidates.

Flat, parabolic, and Fresnel lens systems are contenders for solar thermal concentrators of various power requirements. The emphasis in this report is on glass for parabolic reflective surfaces. The technology status and experimental data on glass, metallic and polymeric concentrators are presented.

CONTENTS

| | | |
|-------|---|-----|
| I. | INTRODUCTION ----- | 1-1 |
| II. | GLASS ----- | 2-1 |
| | A. GENERAL ----- | 2-1 |
| | B. PHYSICAL PROPERTIES ----- | 2-1 |
| | C. SOLAR CONCENTRATOR MIRRORS ----- | 2-2 |
| III. | PERFORMANCE OF REFLECTIVE SURFACES ----- | 3-1 |
| | A. EFFICIENCY OF REFLECTION OF THE SOLAR SPECTRUM ----- | 3-1 |
| | B. GLASS CUTTING AND FORMING ----- | 3-3 |
| | C. DURABILITY ----- | 3-3 |
| | D. METALLIZATION ----- | 3-4 |
| IV. | ENVIRONMENTAL DEGRADATION EFFECTS ----- | 4-1 |
| V. | CONCLUSIONS ----- | 5-1 |
| VI. | FIGURES ----- | 6-1 |
| VII. | TABLES ----- | 7-1 |
| VIII. | REFERENCES ----- | 8-1 |

SECTION VI

FIGURES

| | | |
|-----|---|------|
| 1. | Glass Structure ----- | 6-1 |
| 2. | Classification of Optical Glasses ----- | 6-2 |
| 3. | Simplified Sketch of Two Types of Solar Concentrators ----- | 6-3 |
| 4. | Section of a Conceptual Glass Solar Concentrator Mirror System ----- | 6-3 |
| 5. | Stress Time Characteristics of Glass ----- | 6-4 |
| 6. | Young's Modulus of Various Glasses ----- | 6-4 |
| 7. | Shear Modulus of 0211 Microglass as a Function of Temperature ----- | 6-5 |
| 8. | Poisson's Ratio as a Function of Temperature; 0211 Microglass ----- | 6-5 |
| 9. | Instantaneous Coefficient of Thermal Expansion of 0211 Microglass as a Function of Temperature ----- | 6-6 |
| 10. | Modulus of Elasticity as a Function of Temperature for 0211 Microglass ----- | 6-6 |
| 11. | Transmission Curves of Selected Glasses ----- | 6-7 |
| 12. | Transmittances of Two Glasses in UV, Visible and Near-Infrared ----- | 6-7 |
| 13. | The External Transmittances of Fused Silica Corning No. 7940 and U.L.E. Modified Fused Silica Corning No. 7971, Both Thicknesses 10 mm ----- | 6-8 |
| 14. | The External Transmittances of Corning No. 0160 and No. 7905 (Vycor) Glasses, as Compared With Common Samples of Fused Quartz and Sapphire, all Thicknesses 2 mm ----- | 6-8 |
| 15. | The External Transmittances of Several Samples of Corning and Amersil Glasses ----- | 6-9 |
| 16. | Viscosity - Temperature Curves ----- | 6-10 |
| 17. | Specific Volume of Glass as Related to Temperature --- | 6-10 |
| 18. | Mean Specific Heat of Glasses ----- | 6-11 |
| 19. | Thermal Conductivity of Glasses ----- | 6-11 |

| | | |
|-----|---|------|
| 20. | Thermal Expansion of Three Glass Ceramics ----- | 6-12 |
| 21. | Expansion-Temperature Curves for Typical Glasses ----- | 6-12 |
| 22. | Thermal Expansion of Three Glasses ----- | 6-13 |
| 23. | Thermal Conductivity vs. Temperature ----- | 6-13 |
| 24. | Solar Reflectance ----- | 6-14 |
| 25. | Reflectance Properties ----- | 6-15 |
| 26. | Glass Transmission and Reflection Efficiencies ----- | 6-16 |
| 27. | Reflectivity vs. Wavelength for Fusion Low Iron Glass ----- | 6-17 |
| 28. | Percent Beam vs. Angular Deflection ----- | 6-18 |
| 29. | Angular Deviation From Specular Direction for First and Second Surface Reflections ----- | 6-19 |
| 30. | Schematic of Proposed 3D Vacuum Forming Technique ----- | 6-19 |
| 31. | Reflectivity of Some Common Metals for Normal Incidence as a Function of Wavelength ----- | 6-20 |
| 32. | Degradation of Reflectance on Clean Aluminum Mirrors vs. Wavelength ----- | 6-21 |
| 33. | Glass Mirrors: Specular Reflectance vs. Environmental Exposure Time Location - Albuquerque, NM - 1978 ----- | 6-22 |
| 34. | Percent Reflection Efficiency vs. Wavelength ----- | 6-22 |
| 35. | Cleaning Effects ----- | 6-23 |

SECTION VII

TABLES

| | | |
|-----|---|------|
| 1. | Analysis and Properties of Representative Glasses ---- | 7-1 |
| 2. | Glass Physical Properties ----- | 7-1 |
| 3. | Modulus of Rupture Design Values ----- | 7-2 |
| 4. | Temperature Characteristics of Glasses ----- | 7-2 |
| 5. | Coefficients of Thermal Expansion ----- | 7-3 |
| 6. | Refraction Indices of Glass - Relative to Air ----- | 7-3 |
| 7. | Refractive Indices vs Temperature for Optical Grade Glass Corning No. 7913 (Vycor)----- | 7-4 |
| 8. | Temperature Coefficients of Refractive Index ----- | 7-4 |
| 9. | ASTM Tests Pertaining to Glass ----- | 7-5 |
| 10. | Summary of Reflective Surfaces for Solar Concentrator- | 7-6 |
| 11. | Glass Survey Summary ----- | 7-8 |
| 12. | Reflectance Efficiency of Glass Mirrors at AM2 ----- | 7-9 |
| 13. | Solar Transmission Efficiency of Some Coatings ----- | 7-11 |
| 14. | Optical Properties of Metals ----- | 7-12 |
| 15. | Solar Efficiencies of Silvering Processes ----- | 7-13 |
| 16. | Solar Efficiency versus Silver Thickness ----- | 7-14 |
| 17. | Properties of Coated Chemically Deposited Silver Mirrors ----- | 7-16 |
| 18. | Reflectance Degradation of Desert Mirrors Exposed for 184 days ----- | 7-17 |
| 19. | Reflectance Degradation as a Function of Wavelength for Desert Soiled Mirrors ----- | 7-18 |
| 20. | Washing of Model Heliostat Using Different Washing Solutions ----- | 7-19 |
| 21. | Washing of SRE Heliostats Using McGean Chemical's Washing Solution ----- | 7-20 |
| 22. | Summary Table -- Identification of Reflective Surfaces for Solar Concentrator Applications ----- | 7-21 |

SECTION I

INTRODUCTION

With the increasing interest in the conversion of the sun's energy to do useful work for mankind, the solar-thermal power system is attractive and has been for considerable time. For some of the historical aspects, see References 1 through 6. The use of reflective surfaces, i.e., mirrors, for redirection and concentration of the solar light rays is becoming more important. This report treats the properties of basic glass materials in general and glass of second-surface mirrors in particular. These subjects are vast and are extensively documented in the literature. In this report, however, the emphasis is on the glass and mirror technology useful to the designer of high performance solar concentrators.

Primary consideration is given to high reflectance glass mirrors with a secondary emphasis on coated metal reflectors. The approach is to survey the most promising candidates from a historical data base, present status, and future capability. The aspects treated include efficiency of solar reflectances, effect of dust and contamination accumulation, stability of reflective metallization, environmental effects, cleaning effects, and projected price structure.

Because of the large number of figures and tables in this report they are listed separately at the end to minimize searching and to facilitate their use.

SECTION II

GLASS

A. GENERAL

There are many different types of glass that are available for solar concentrator reflection surfaces. Over 100,000 types of glass are known, and probably 750 types are produced commercially. In the selection of the specific glass surfaces to be used various factors should be considered, such as optical and mechanical properties, durability, resistance to hail, rain and other environmental influences, and thermal expansion.

Categorizing glasses by their method of production, the majority are fusion, float or sheet (drawn) glass. By chemical composition, the primary types are the following:

- (1) Soda-lime
- (2) Lead-alkali
- (3) Borosilicate
- (4) Aluminosilicate
- (5) Fused silica

The elemental composition of common glasses is shown in Table 1. The borosilicates and aluminosilicates are found to be superior for solar concentrator applications. The character of these and other types of glasses are treated in the following section.

B. PHYSICAL PROPERTIES

Glass has been defined as an inorganic product of fusion which has cooled to a rigid condition without crystallizing (Reference 7). Thus, glass is a rigid fluid. But some organic rigid liquids can be properly called organic glass; therefore, glass should not be limited to inorganic materials. However, all glasses referred to commercially as glass are inorganic. The typical atomic structure is shown in Figure 1b.

The general physical properties of typical glasses applicable to solar concentrator designs are summarized in this report. The wide variability of the data is apparent and the physical properties of glass composition are complex (Figure 2). The two best references for the solar glass designer are Strand (Reference 8) and Corning Glass Work's Properties of Glasses and Glass-Ceramics (Reference 7), although many other fine treatises exist. See References 9 through 67.

General engineering data on the characteristics of glass for solar thermal concentrators are summarized in Tables 2 through 8 and Figures 5 through 23. Mechanical, data transmittance, index of refraction, viscosity, and thermal data are given. For further details concerning the specific parameters, the reader is referred to the reference indicated. Some of the more important tests pertaining to glass, taken from ASTM literature, are shown in Table 9. Other documents and reports containing information on high solar transmittance glass are References 68 through 110.

C. SOLAR CONCENTRATOR MIRRORS

The general concepts for concentrating solar energy are sketched in Figure 3a for the parabolic mirror system and in Figure 3b for the more advanced Fresnel lens concentrator. A simplified drawing of a section of a conceptual glass mirror system, the primary subject of this report, is shown in Figure 4. A mirror may consist of many more layers than those shown. For example, the inner protective coating and the sealant may both consist of multiple layers. Commercial mirrors usually have different types of metallic coatings, sealants, and protective over-coatings which vary in number and thicknesses.

Major problems exist in translating the conceptual mirror designs into durable, low cost, high reflectance systems that will withstand the environmental elements over the projected long life of 20 or more years.

The physics of the light reflection from a typical second-surface glass mirror, in simplified form, is shown in Figure 24a. Multiple reflections are not shown. The incident light ray undergoes spreading due to reflection, this is known as the specular component of the reflected light, while the electromagnetic radiation reflected and emitted from the glass surface over the remaining part of the hemisphere is known as the diffuse component. The objective in the fabrication of a high performance glass mirror system is to concentrate as much of the light beam as possible into the specular component, and hence into the cavity receiver for eventual conversion into power.

For comparison, a front-surface metallic mirror of aluminum is shown in Figure 24b. Coated polymeric surfaces are not shown, but may become future candidates as the technology evolves.

SECTION III

PERFORMANCE OF REFLECTIVE SURFACES

A. EFFICIENCY OF REFLECTION OF THE SOLAR SPECTRUM

The investigation of the efficiency of solar concentrators includes data from in-house JPL measurements, Sandia Corp., McDonnell Douglas Astronautics Company-West and other vendor supplied information. These data have been tabulated along with explanatory notes, when applicable, in Tables 10 through 13.

High reflectance, in the 80-96 percent range, is needed for an efficient solar mirror surface, and the higher the better. From the data in Tables 10 and 11, it is seen that when silvered Corning fusion glass 0317, Schott B270 and fusion glass modified 7806 (low ferrous content) are determined to have superior reflectance. The modified 7806 glass is yet to be put into production; however, due to the nature of the fusion process, it is expected to be satisfactory when available in limited quantities next year.

The greenish color imparted to glass by the presence of iron is highly undesirable (Figure 25) and ways have been investigated in the glass industry to overcome the effect of its absorption. From purely economic considerations, the removal of iron has certain practical limits. At present, a number of approaches appear successful, including the addition of flourides and phosphates to counter the effect (Reference 88). Another approach is to use the fusion process in making the glass. In this process, the iron remains in the glass but is converted largely to the ferric state (96% Fe₂O₃, 4% FeO) and the infrared absorption effect is significantly reduced. The physics and chemistry of the role of iron in glasses is complex and beyond the scope of this report.

Total solar hemispherical measurements of 0317 fusion glass shown in Table 10 (item 29) were performed at JPL using a Model MS 250, Gier Dunkle reflectometer (Gier Dunkle Instrument Inc., Santa Monica, CA). A total hemispherical reflectance of 93 percent was obtained. Other measurements show values as high as 95 percent. Recent mirror reflectance data measured at a single wavelength (500 nanometers) by Sandia Corp. are included in the table. Where the coefficients are measured at other frequencies, they are given in parentheses below the coefficient.

Four generalized coefficients, R_1 and R_2 , and the standard deviation parameters, σ_1 and σ_2 , can be utilized to describe the reflectance of solar energy from a given material. See Reference 98. Assuming the reflectance profile can be described as the sum of two normal distributions, the equation is:

$$R(\Delta\theta) \propto R_1 \exp \left[\frac{-\Delta\theta^2}{2\sigma_1^2} \right] + R_2 \exp \left[\frac{-\Delta\theta^2}{2\sigma_2^2} \right]$$

where R is the reflectance and $\Delta\theta$ is the deviation of the reflected beam from the specular direction. This equation gives remarkable fit to the measured data. Where known, these coefficients are shown in the tables. The general theoretical relationship between reflection and transmission for glass is shown in Figure 26 along with the recent data for thin Corning fusion glass 0317 with good metallization. The metallization aspects are treated later in this report.

Recent solar reflectance data of Sandia Laboratories, for thin solar concentrator glass of 0.058 inch thickness, is given in Figure 27. The angular deflection of light transmission has been compiled in Figure 28.

Aluminosilicate glasses, such as 0317 fusion, have excellent optical flatness for glasses as thin as 2.794 mm (0.110 inch) to 2.287 mm (0.090 inch). However, for very thin glass, these data show that optical flatness is not as good with 46 percent of the transmitted beam power outside 1 milliradian and 22 percent for 2 milliradians half-angle respectively (Reference 96). Similar types of measurements are needed for mirrors.

Angular deviations from the first and second surfaces, for various float glasses after light incident at 45° , are shown in Figure 29 (Reference 96). In general, other data indicates the with draw direction to be smoother than the across draw direction. Similar measurements for other reflection angles are needed.

It is important to match the thermal coefficient of expansion over the postulated service temperature (-30° to $+150^\circ\text{F}$). Where available these coefficients are shown in Table 10 for high reflectivity glasses.

For front-surface coated aluminum reflectors, Alcoa's Type S460667 is reported to have a total hemispherical reflectance of 92 percent, while the Kingston Industries, Inc., Kingflux, exhibits about 85 percent. JPL measurements of Kingflux aluminum found 85.4 percent. An exposure experience of approximately 15 years of shipboard applications indicates considerable durability for this latter surface.

Preliminary price information on glass and aluminum reflective surfaces has been accumulated in Tables 10 and 11. Initial estimates show glass to be cheaper than front-surface aluminum. Solar mirror panel quality glass in relatively large quantities, the order of one million square feet, may be obtained in the near future for 0.65 - 0.80 dollar per square foot. The silver metallization in mass production may be obtainable in the range of 0.5 - 1.00 dollar per square foot of additional cost. Although the front-surface mirrors are more expensive at the present time, this situation could change rapidly as the solar concentrator technology matures.

B. GLASS CUTTING AND FORMING

In order to successfully fabricate a solar concentrator, cutting and 3-dimensional surface forming must be undertaken. These are very formidable problems for the thin glass needed for solar concentrators. Research is required to perform these operations on large sheets with low breakage. Corning fusion glass, type 0317, in thicknesses of 1.5 mm (0.060 inch) and 1.02 mm (0.040 inch) is difficult to cut, according to JPL experience. One conceptual possibility for cutting is laser melting.

Three dimensional cold forming has been proposed as a method to fabricate the parabolic curved surface, and this method may be feasible for thin glass. Simplified, this is shown schematically in Figure 30. Basically, the thin glass sheet is pulled against the substrate, in this case cellular glass with an adhesive bond surface. Air pressure forces the plastic cover against the glass and this configuration is held until the glass-substrate interface bond is formed.

C. DURABILITY

Of the surfaces studied, glass appears to have the most durable characteristics to withstand low humidity environments. Egyptian glasses over 2,000 years old are strong evidence for its stability. Atmospheric water vapor readily attacks glass as do other environmental components such as hail and sand. Cleaning solutions also attack glass. On the basis of the available data, glass appears to be very resistant to environmental forces in general. However, the durability of the underlying metallic coating is of prime importance for obtaining high reflectivity for extended times.

A possible method for assessing the aging effects on mirrors exposed to the natural environment would be the evaluation and testing of automobile side-view mirrors. This methodology is used in Reference 121. At least two manufacturers, General Motors (Buick) and Mercedes-Benz, are reported to have produced high quality silvered side mirrors for many years. These two manufacturers maintained a strict quality control for their outside mirrors while most other companies did not; moreover, most other manufacturers use mirrors with less reflective materials than silver. Buick's experience in quality mirrors goes back to the early 1920's.

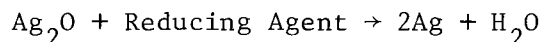
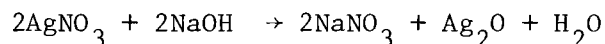
JPL has obtained both types of environmentally aged mirrors including approximately 58-year old Buick mirrors, for evaluation. Of course, the outdoor exposure may be appreciably less than this if the auto was garaged. Therefore, an attempt was made to select for analysis those mirrors with the most corroded mirror fixture. Visual inspection of both the glass and metallization shows that they are in excellent condition. This preliminary evidence indicates that high quality, long-lived mirrors for solar power applications can be successfully designed.

D. METALLIZATION

Second-surface glass mirrors require a high quality, protected metal coating to meet the extended service life for solar concentrators. Some evidence indicates that even higher performance reflective surfaces may be achieved by combinations of metals, such as silver and aluminum, or the use of more innovative techniques.

Silver is the metallic reflective coating that is the most straight-forward to apply and it is commonly used on second-surface glass mirrors for many applications. The chemical method is more economical than vacuum-deposited silver and is most frequently used in commercial mirroring processes.

The silvering reaction is:



Typical protective coatings used in the mirror industry are the following:

- (a) 753.4 mg/m² (70 mg/ft²) of silver
- (b) 161.5 mg/m² (15 mg/ft²) of copper
- (c) 86,114 mg/m² (8 mg/ft²) of mirror sealant

The reflectivity of vacuum deposited silver for 0317 fusion glass has been established by Sandia Corp. to be superior to the chemical process but only results in one percent greater reflectivity over the spectrum (Table 10, items 27 and 28). A solar hemispherical reflectance of 95 ± 1 percent was obtained with the former method. The reflectivity of common base metals is shown in Figure 31 and Table 14. These data show that silver has the highest reflectivity above the ultraviolet range while aluminum is a better reflector in the ultraviolet range. The efficiency of the silvering process is shown in Tables 15 and 16. The effect of oxidation of aluminum reflectivity with time is shown in Figure 32. Further information on reflective surfaces and their degradation due to solar exposure is given in the following section.

SECTION IV

ENVIRONMENTAL DEGRADATION EFFECTS

Degradation of high-performance surfaces can occur due to environmental effects, such as dust, sand, and miscellaneous contamination. Little is known of the long-term effect of these environments on mirror surfaces. However, in general, the effects vary with the type of material, length of exposure, site location, type of coating(s) and cleaning techniques. Preliminary JPL data indicates, however, that bare plastics tend to accumulate dirt with a greater tenacity than glass. A detailed treatment of these subjects is beyond the scope of this summary and is planned to be treated in another report (Reference 105).

The general effects of reflectance degradation for desert-soiled mirrors is shown in Tables 18 and 19 and Figure 33 shows the dust-rain effects in the Sandia Corp. program (Reference 115). A decrease in reflectance of 12% due to dust is noted. Wavelength dependence is shown in Figure 34 (Reference 103). Preliminary information indicates that upside-down stowage for the mirror appears to be better compared to face-up stowage. This is true in spite of the fact that overnight dew or frost may run off and sometimes clean the mirrors during morning deployment in the latter case.

Various cleaning methods are under consideration but further research is required to identify the type of cleaning agent that will minimize damage to the concentrator surface over an extended period of time. The various cleaning procedures are summarized in Tables 20, 21 and a summary of previous glass cleaning effects is shown in Figure 35. At this time, the aluminosilicate and the borosilicate glasses are predicted to be superior on the basis of these tests. (Reference 113).

SECTION V

CONCLUSIONS

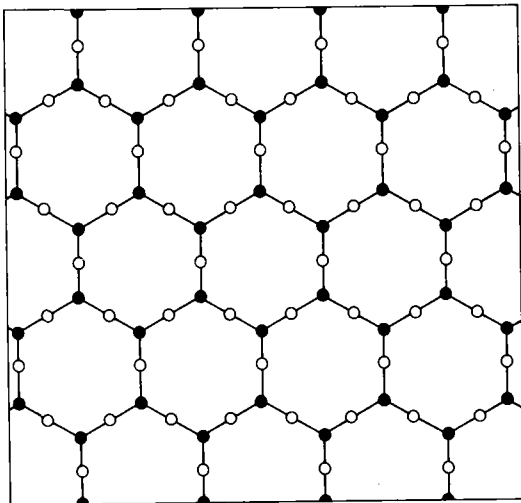
In this survey, four general types of promising reflectors for solar concentrator applications have been identified. They are summarized in Table 22. On the basis of presently available transmission data, Corning fusion glass, type 0317 and Schott glass, type B270, and float glass are recommended for parabolic solar concentrators due to their good potential reflectance, availability, durability, thermal expansion characteristics and projected price structure.

For future applications, however, modified Corning 7806 fusion glass and coated, anodized aluminum surfaces bear watching because of their possible importance. For the Fresnel lens type of solar concentrator, special glass and plastic materials appear to be superior.

Additional test data on the durability of metallic surfaces of high reflectance mirrors are needed in order to prove that environmentally resistant mirror systems can be achieved. Preliminary observations of the good stability characteristics of 58-year old auto mirrors indicate that this can probably be realized.

Although this survey of glass and mirrors treats many engineering parameters, it is by no means exhaustive and should be regarded as preliminary information. Data are frequently based upon a limited number of samples and some metallizations and glasses are not completely characterized. The data are sufficient, however, to indicate the trends.

(a) CRYSTALLINE STRUCTURE



(b) GLASS STRUCTURE (LOW ORDER)

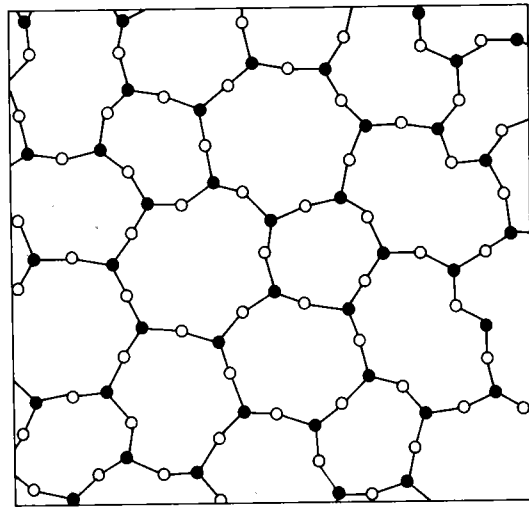


Figure 1. Glass Structure (Ref. 39)

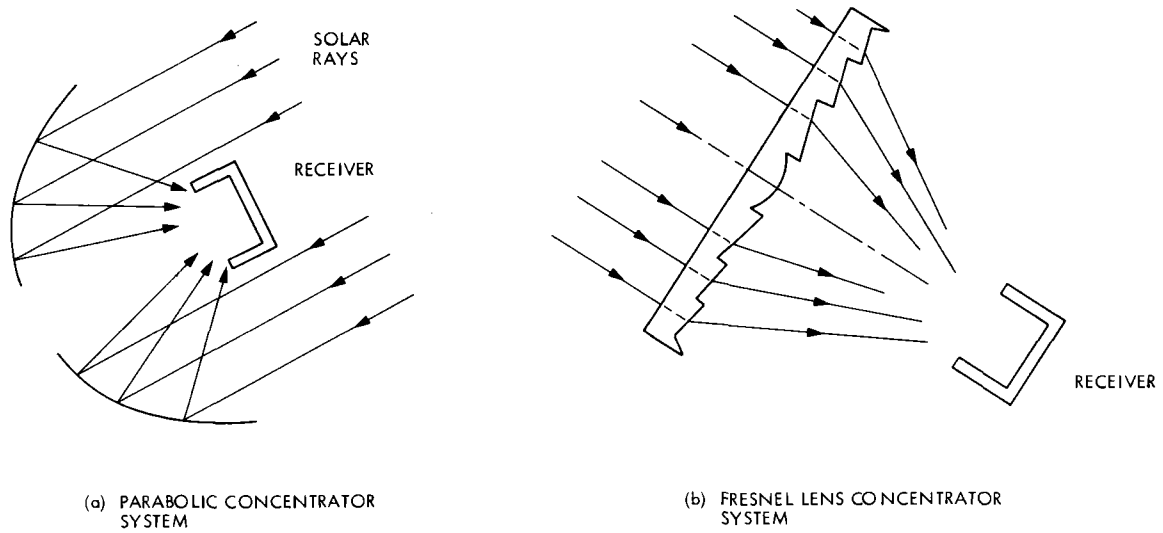


Figure 3. Simplified Sketch of Two Types of Solar Concentrators

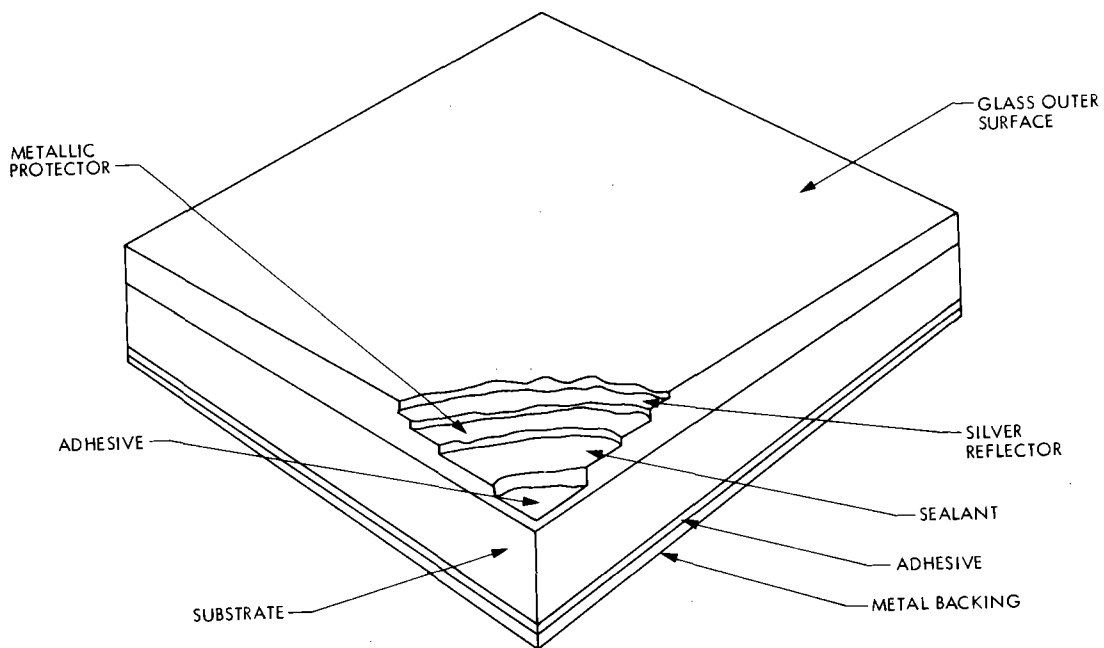


Figure 4. Section of a Conceptual Glass Solar Concentrator Mirror System

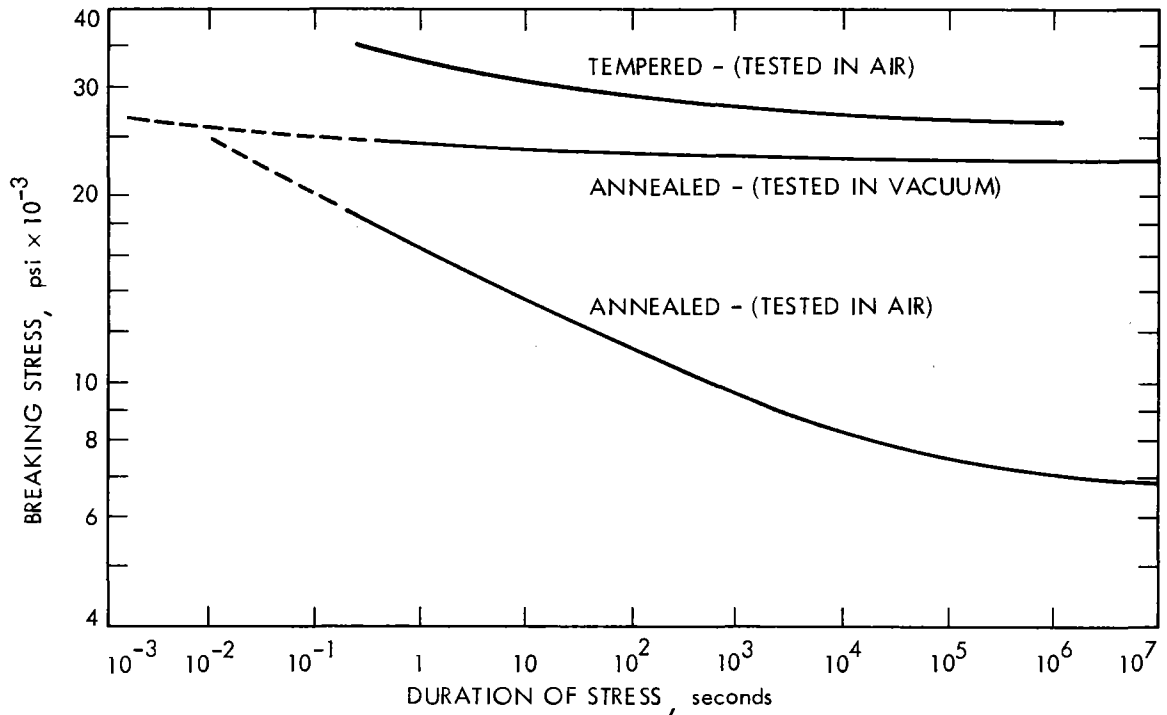


Figure 5. Stress Time Characteristics of Glass (Ref. 36)

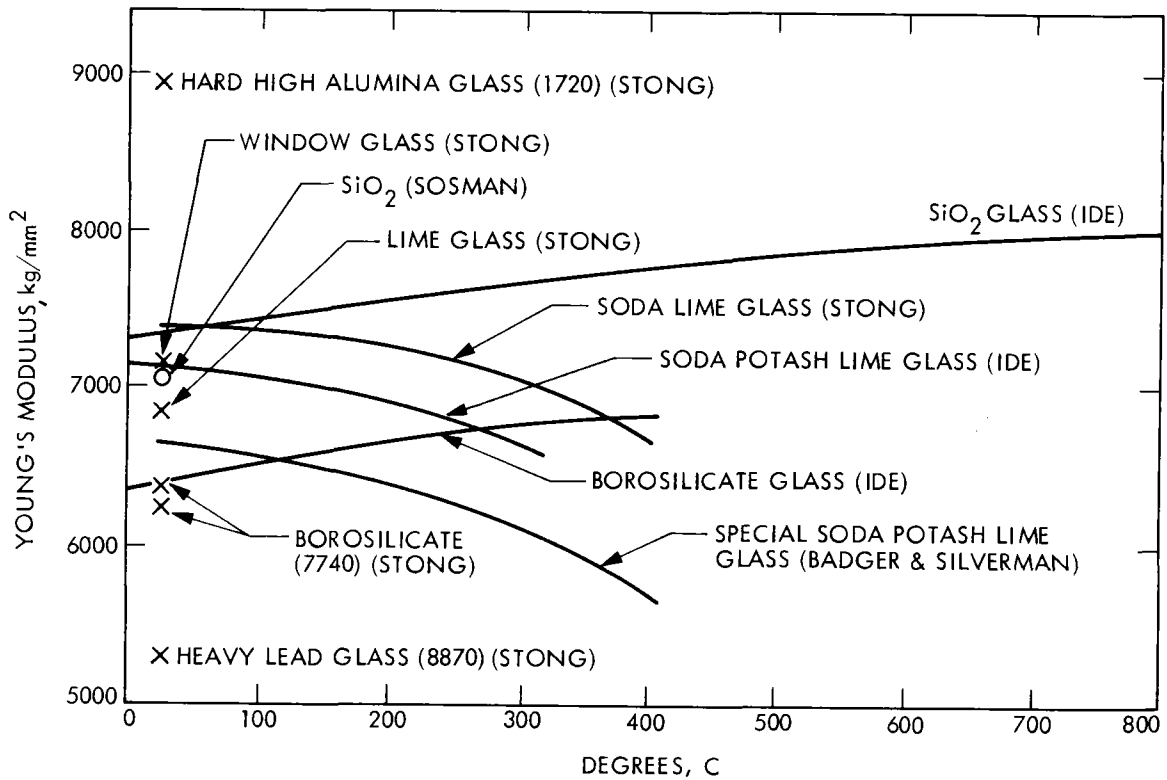


Figure 6. Young's Modulus of Various Glasses (Ref. 36)

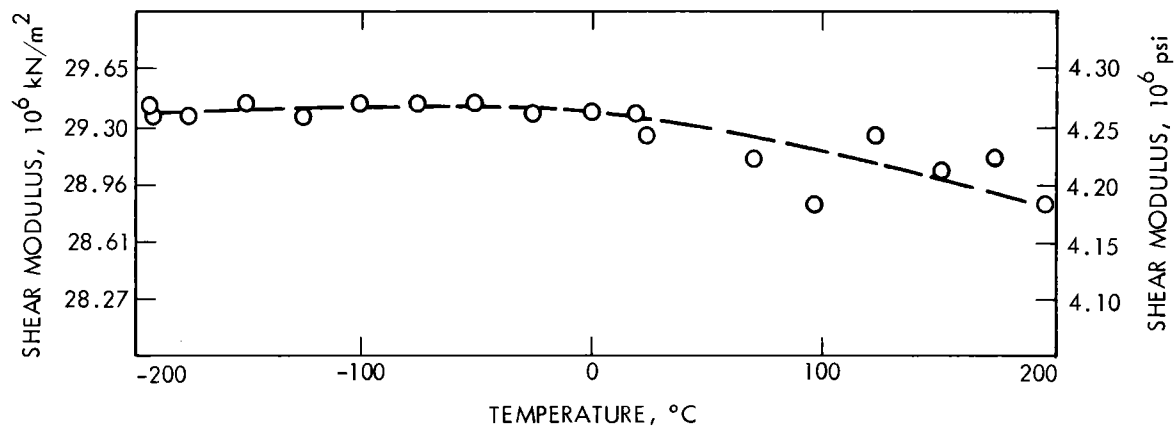


Figure 7. Shear Modulus of 0211 Microglass as a Function of Temperature (Ref. 114)

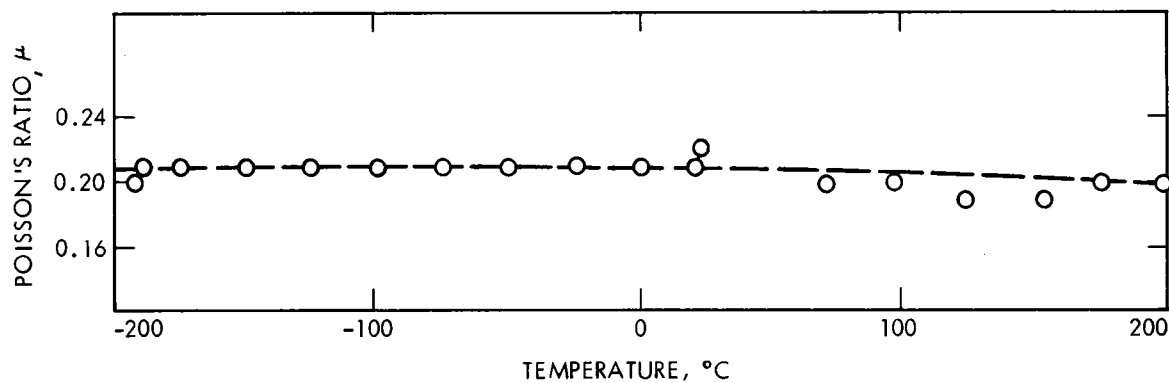


Figure 8. Poisson's Ratio as a Function of Temperature; 0211 Microglass (Ref. 114)

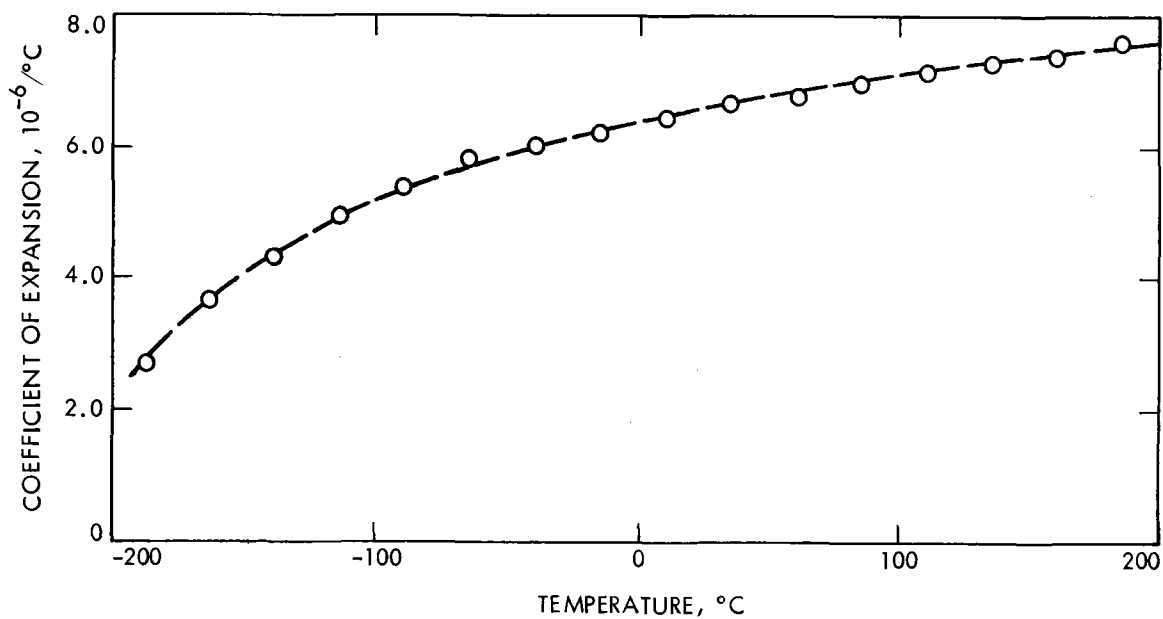


Figure 9. Instantaneous Coefficient of Thermal Expansion of 0211 Microglass as a Function of Temperature (Ref. 114)

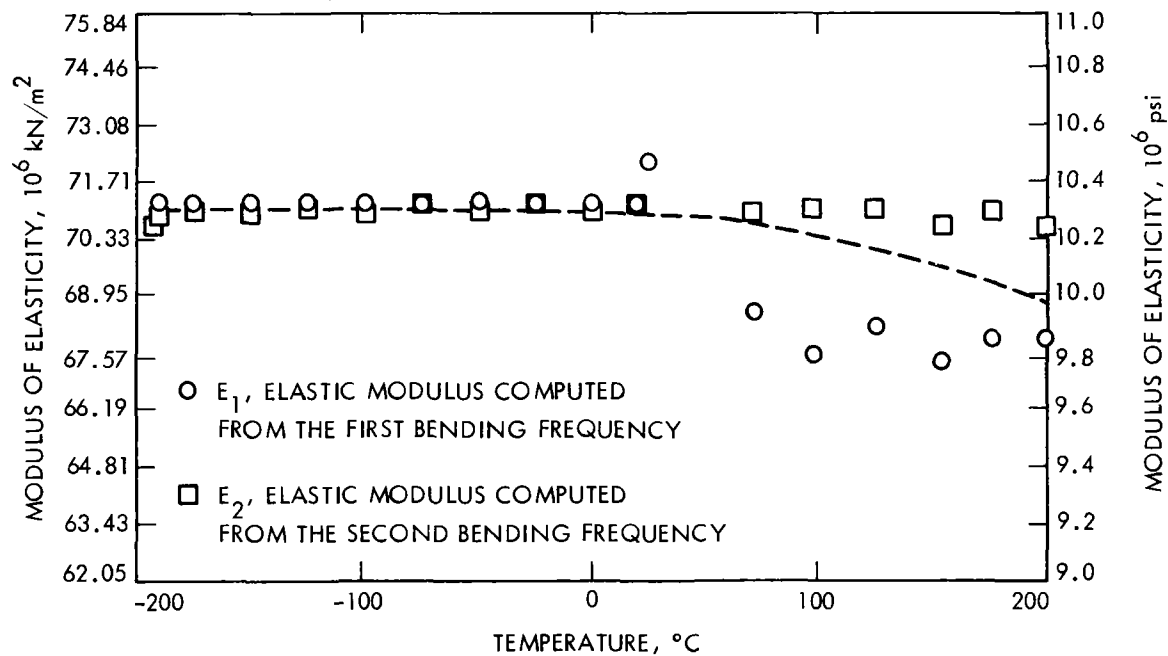


Figure 10. Modulus of Elasticity as a Function of Temperature for 0211 Microglass (Ref. 114)

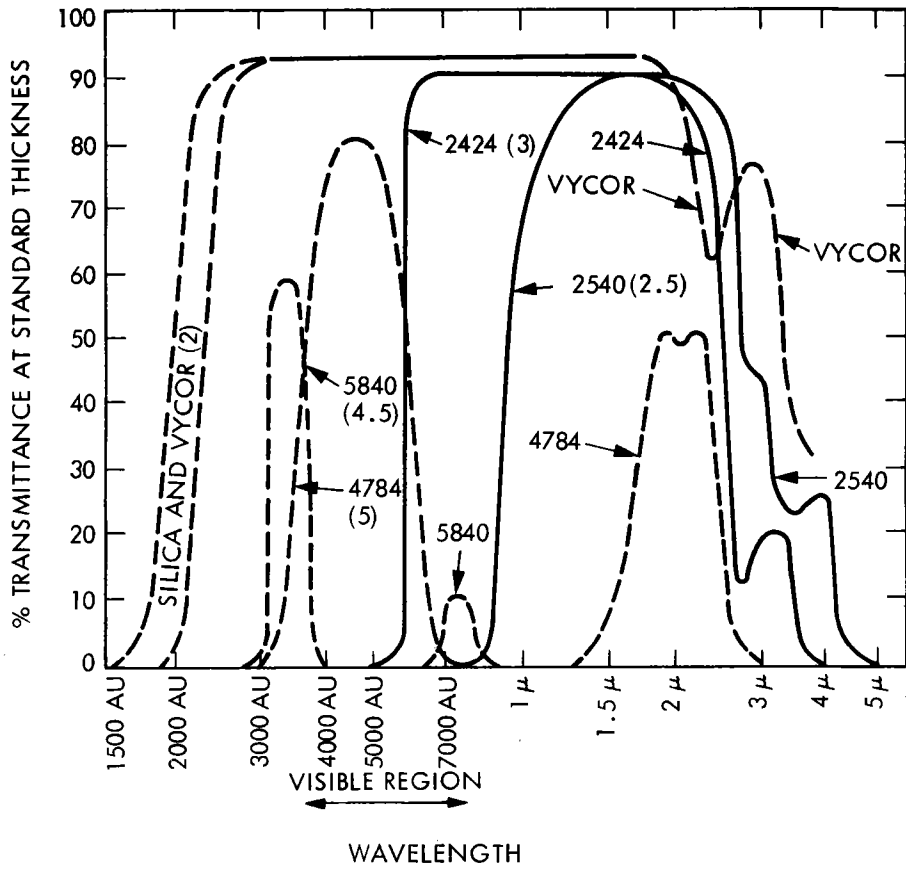


Figure 11. Transmission Curves of Selected Glasses (Ref. 36)

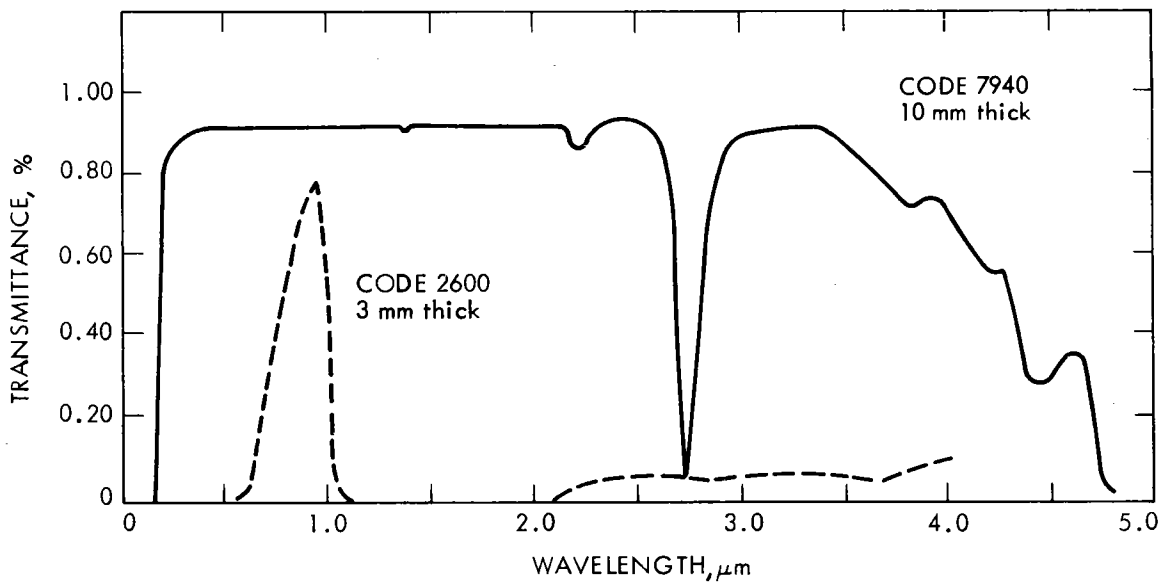


Figure 12. Transmittances of Two Glasses in UV, Visible and Near-Infrared (Ref. 7)

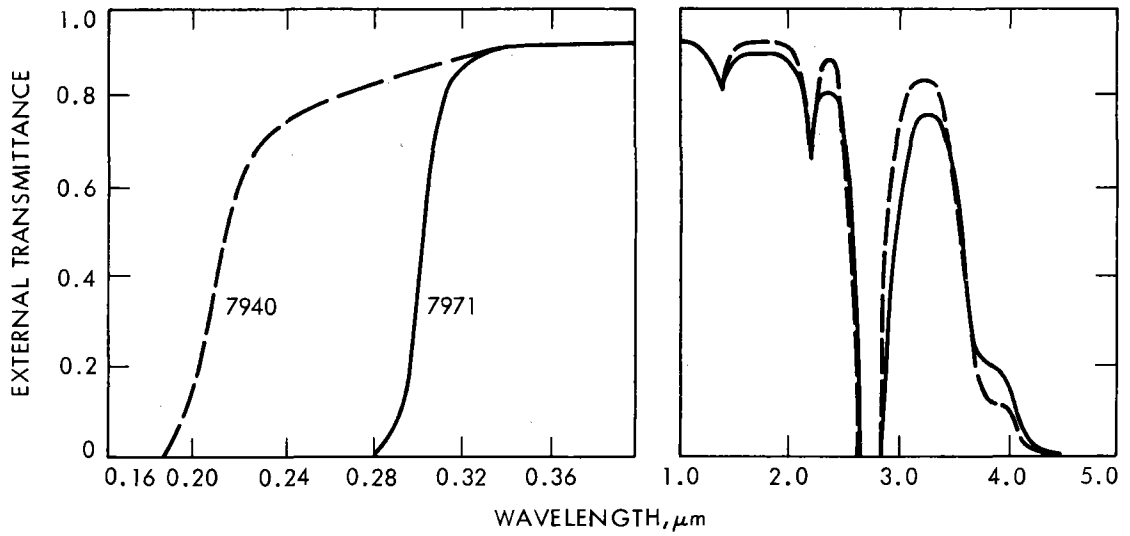


Figure 13. The External Transmittances of Fused Silica Corning No. 7940 and U.L.E. Modified Fused Silica Corning No. 7971, Both Thicknesses 10 mm. (Ref. 53)

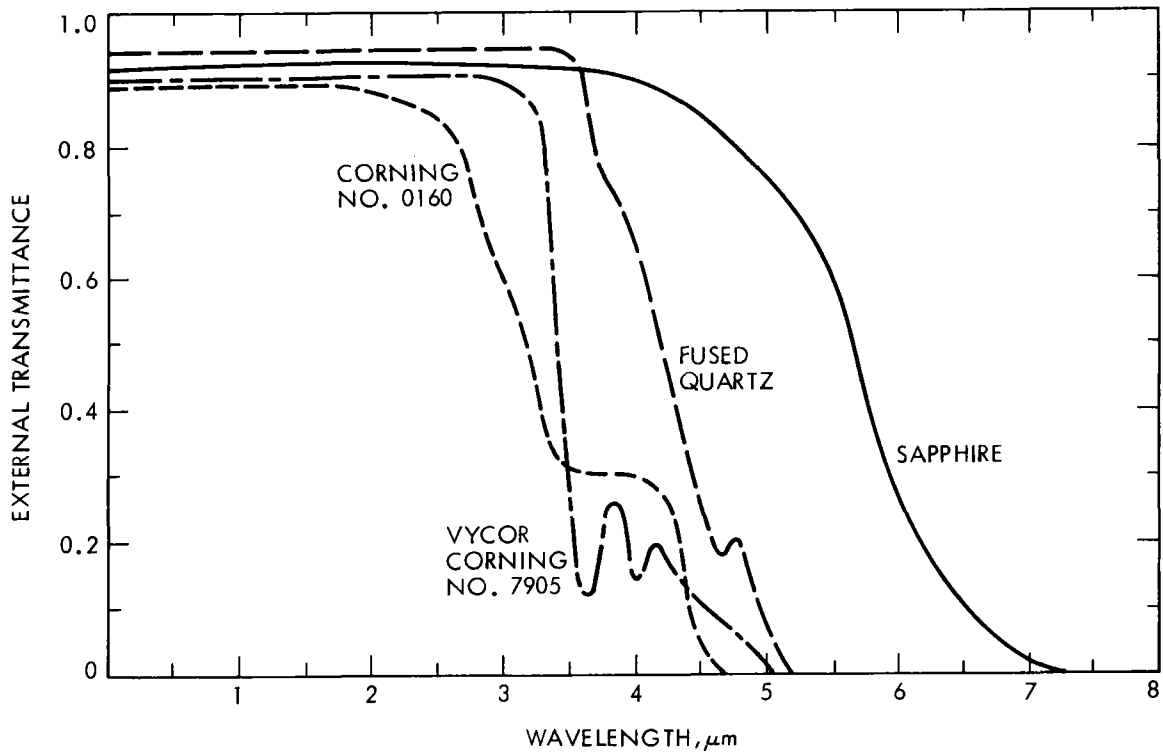


Figure 14. The External Transmittances of Corning No. 0160 and No. 7905 (Vycor) Glasses, as Compared With Common Samples of Fused Quartz and Sapphire, all Thicknesses 2 mm (Ref. 53)

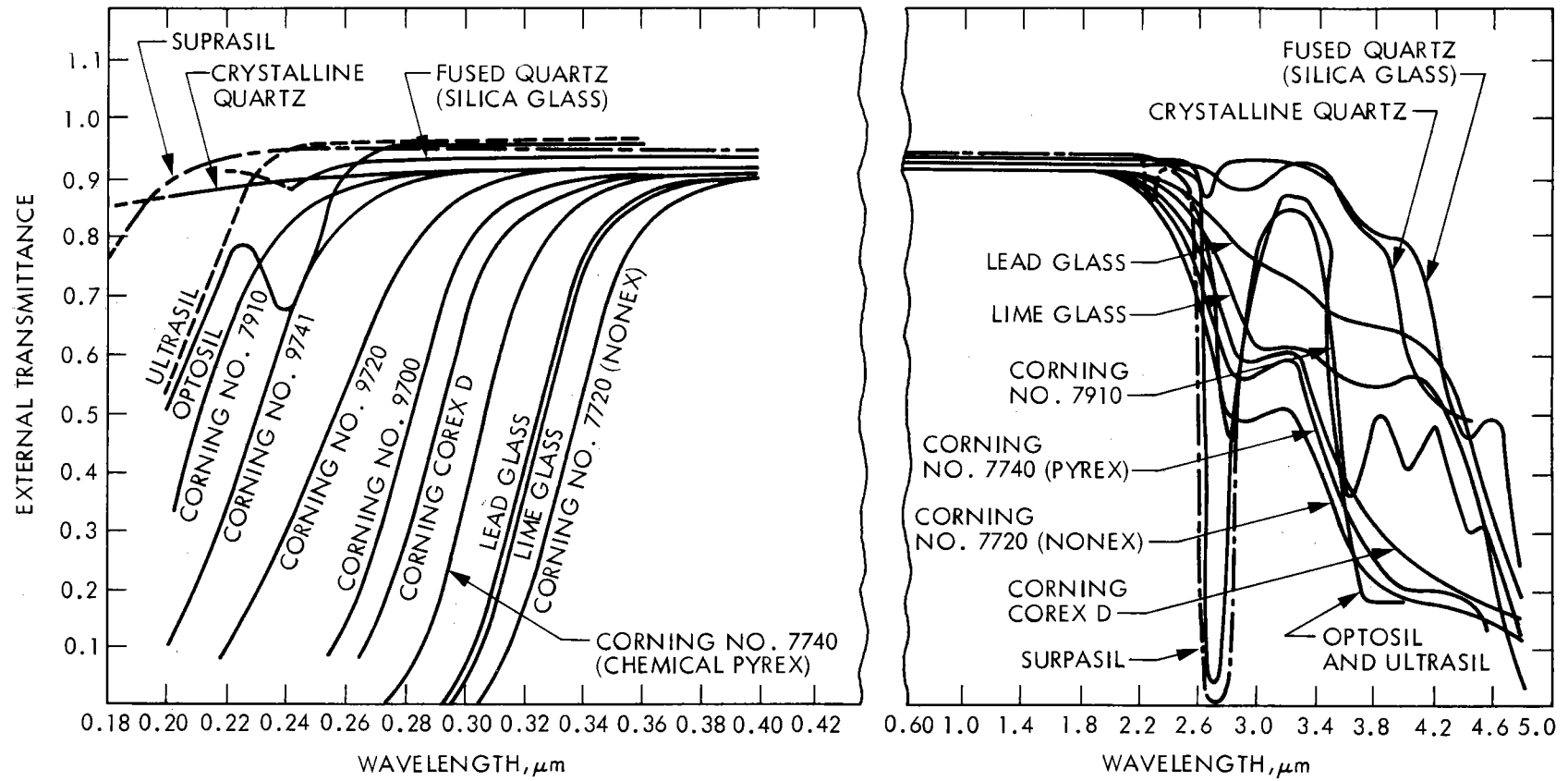


Figure 15. The External Transmittance of Several Samples of Corning and Amersil Glasses. (Ref. 53)

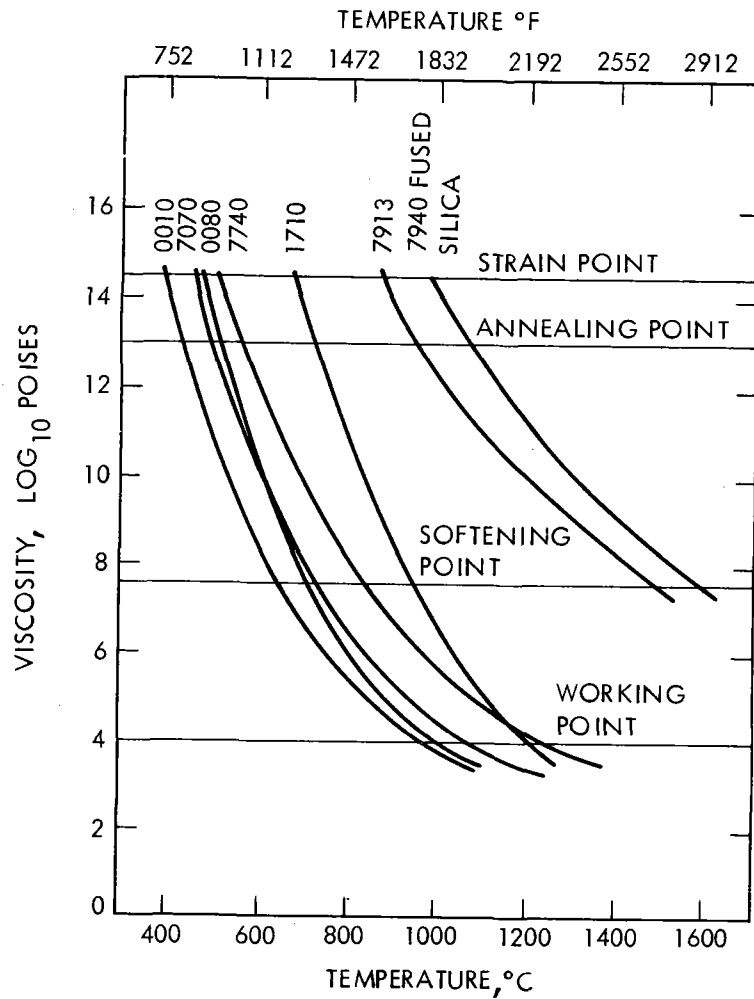


Figure 16. Viscosity - Temperature Curves (Ref. 7)

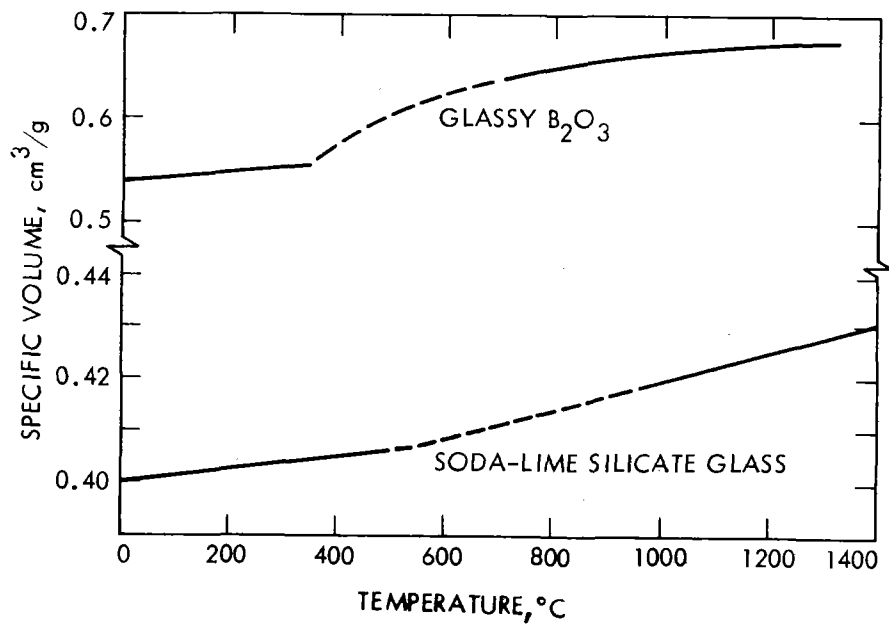


Figure 17. Specific Volume of Glass as Related to Temperature (Ref. 39 and 110)

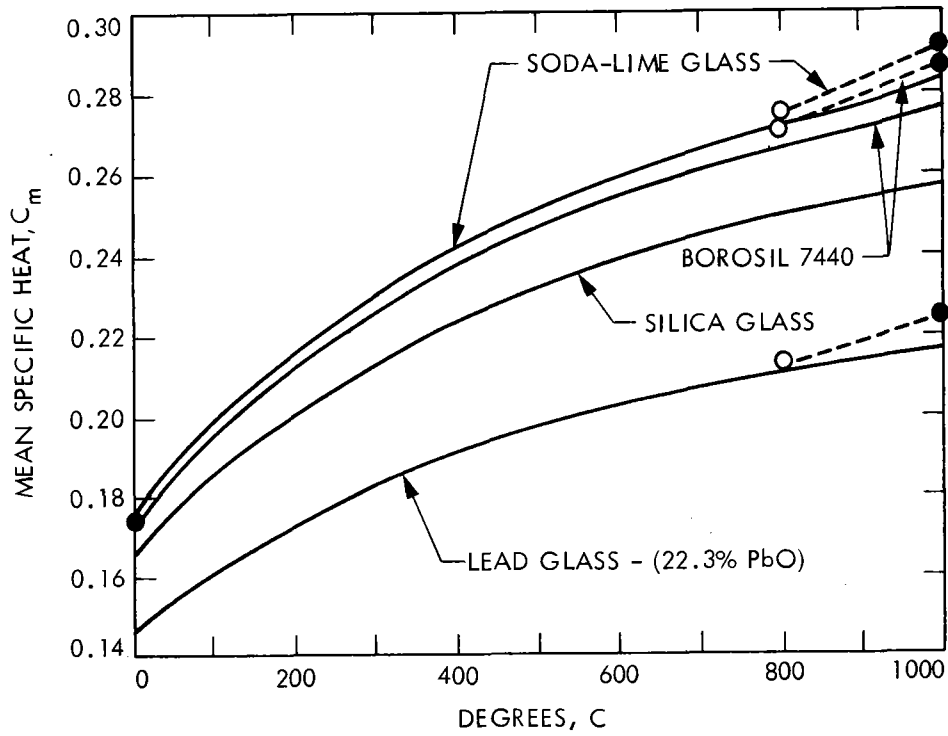


Figure 18. Mean Specific Heat of Glasses (Ref. 36)

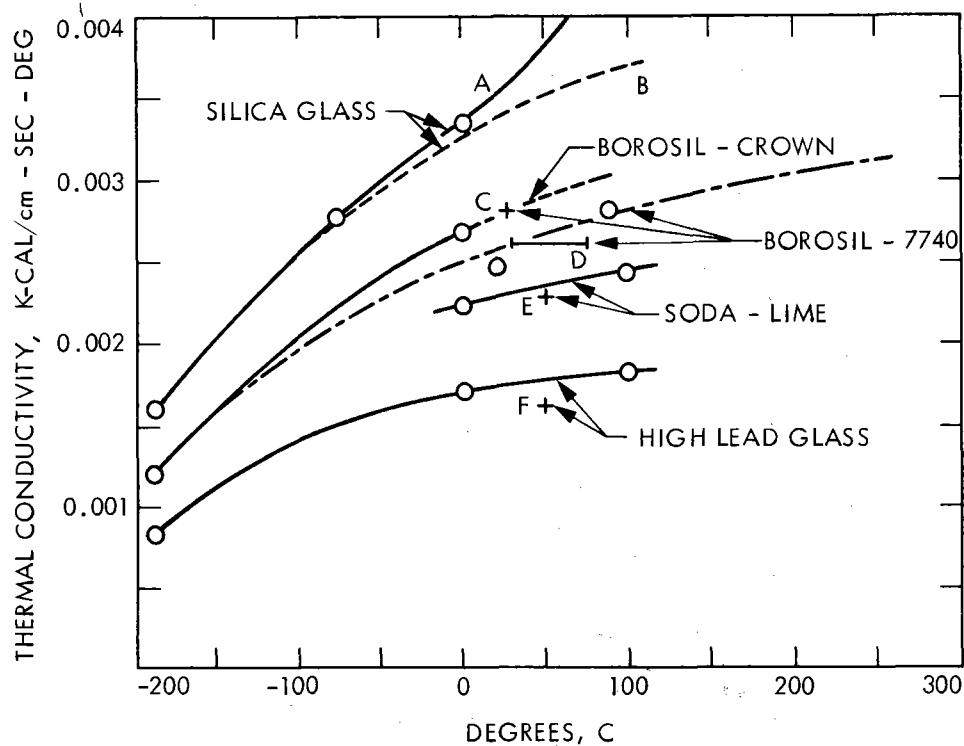


Figure 19. Thermal Conductivity of Glasses (Ref. 36)

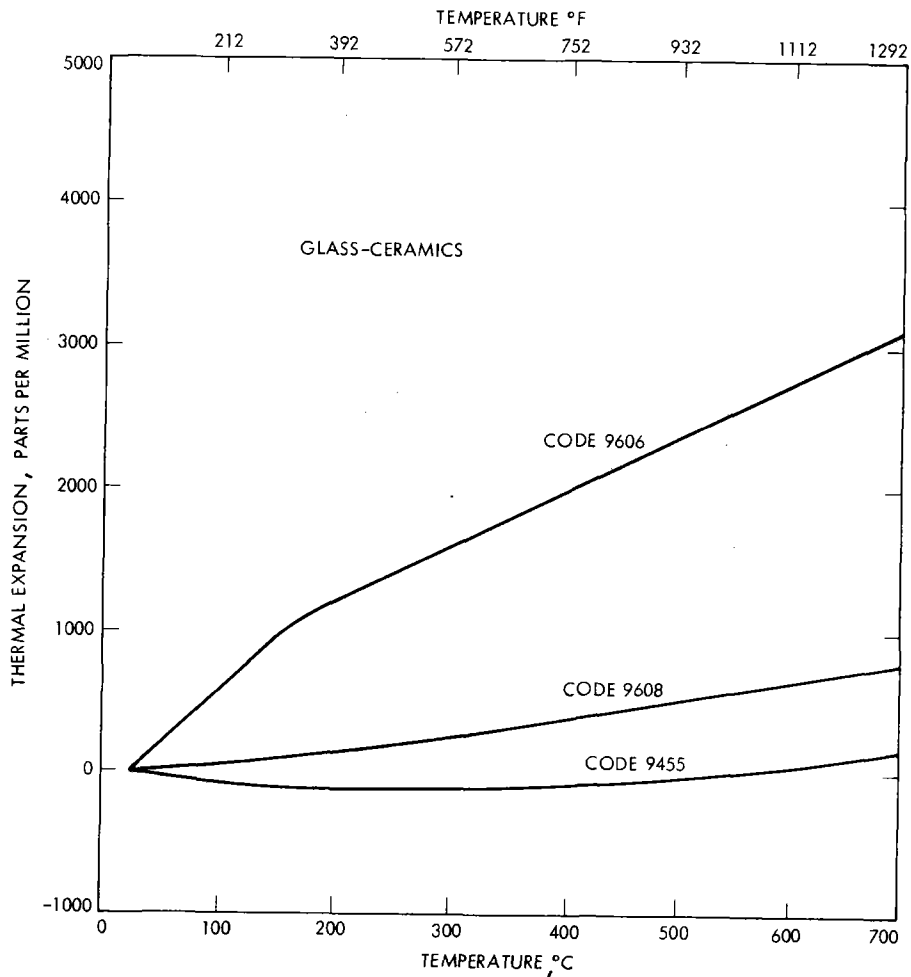


Figure 20. Thermal Expansion of Three Glass Ceramics (Ref. 7)

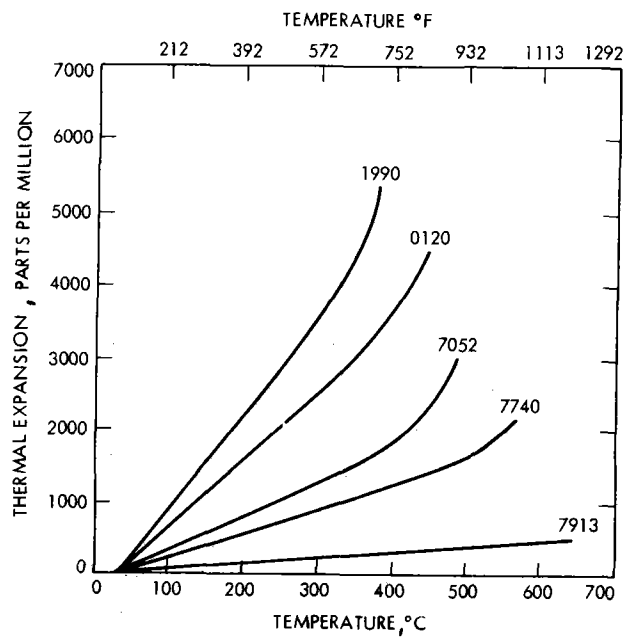


Figure 21. Expansion-Temperature Curves for Typical Glasses (Ref. 7)

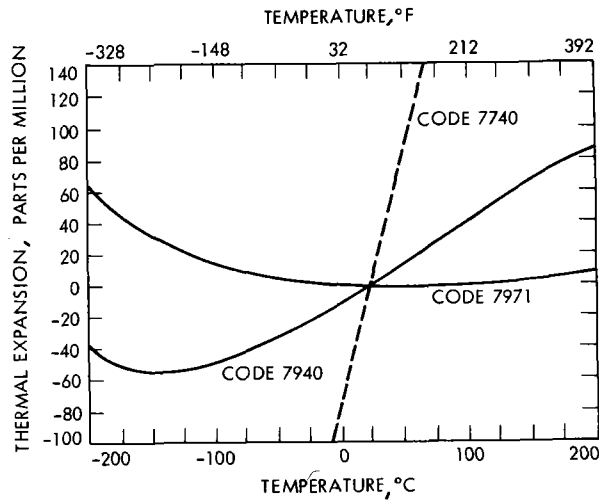


Figure 22. Thermal Expansion of Three Glasses (Ref. 7)

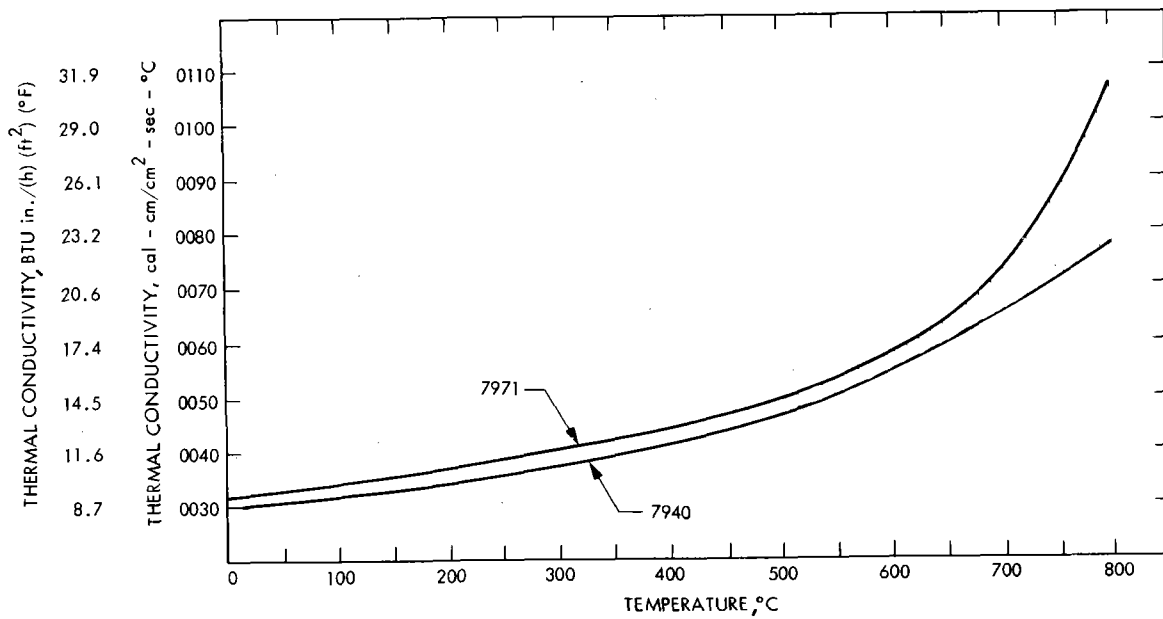
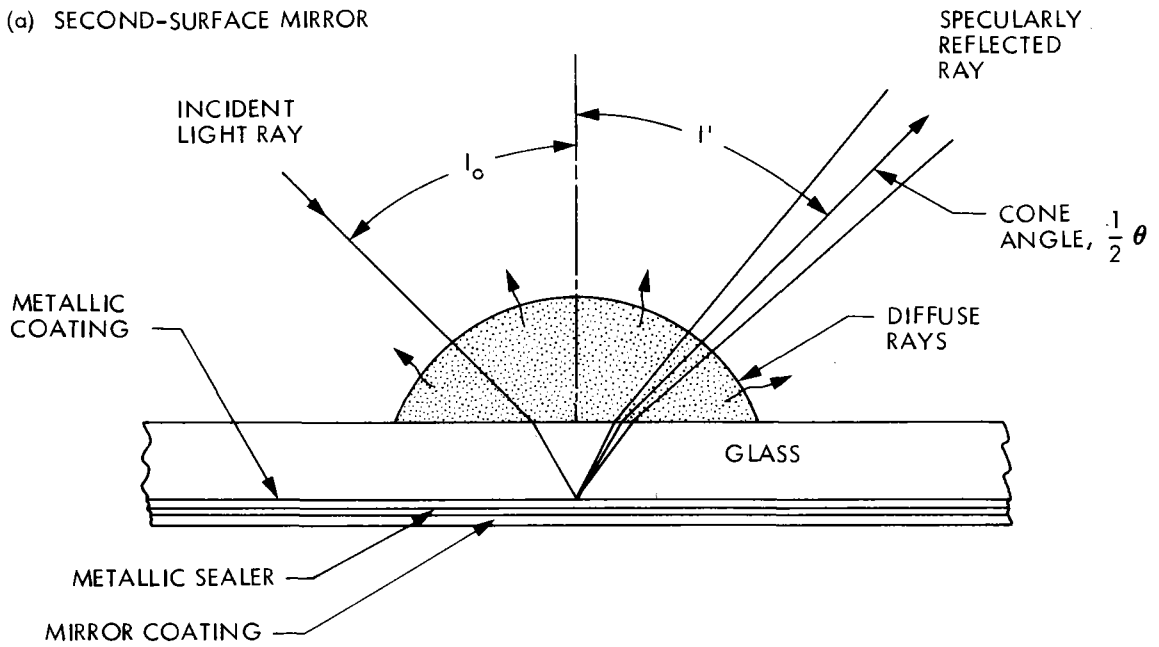


Figure 23. Thermal Conductivity vs. Temperature (Ref. 7)

(a) SECOND-SURFACE MIRROR



(b) FIRST-SURFACE MIRROR

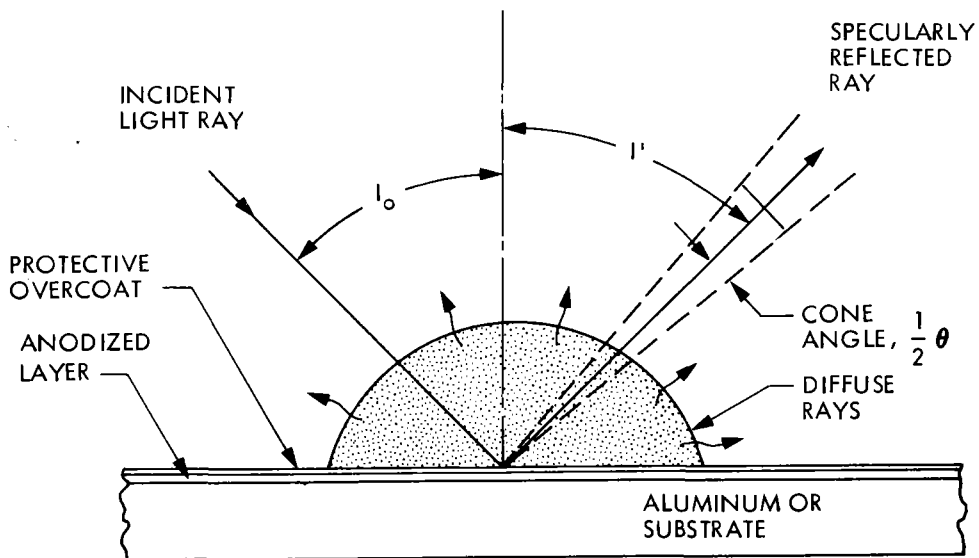
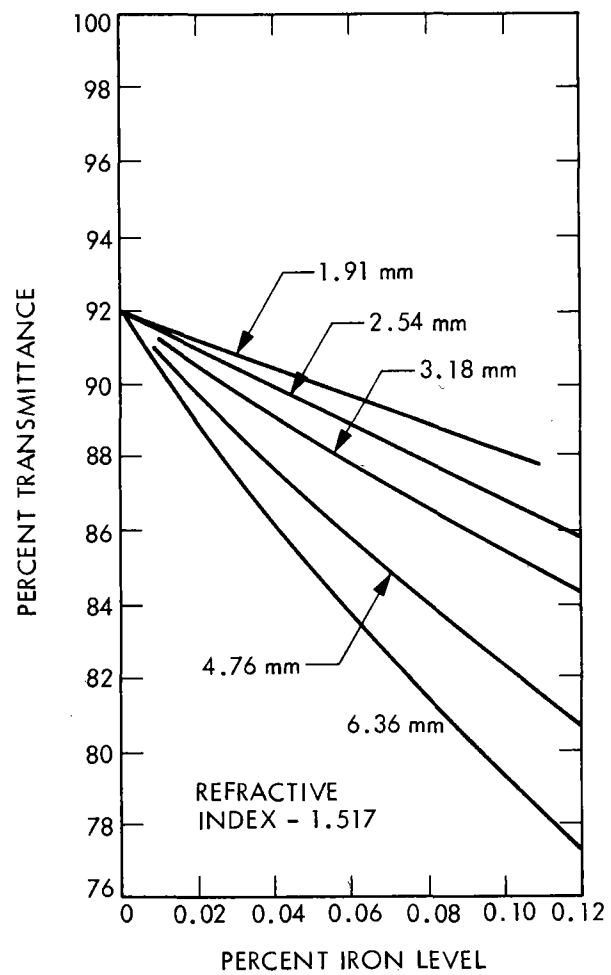


Figure 24. Solar Reflectance

(a)



(b)

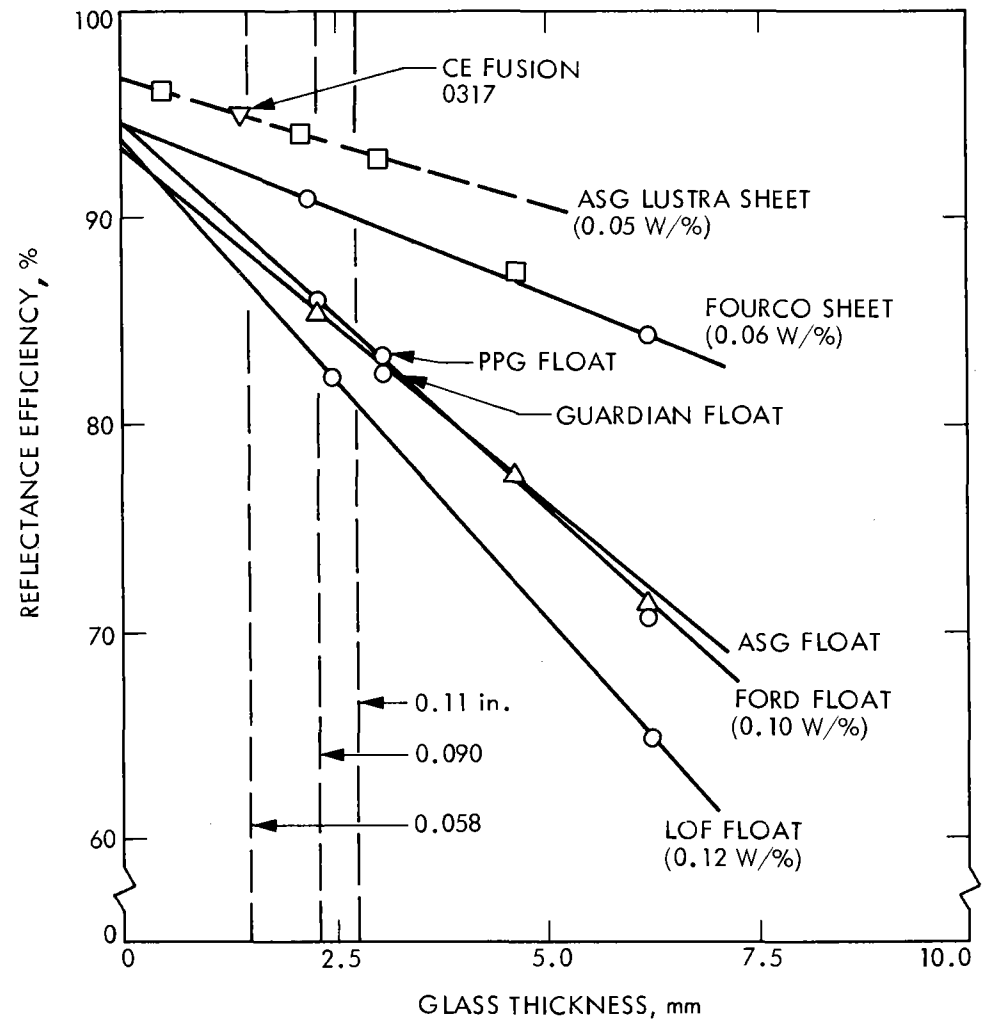


Figure 25. Reflectance Properties (Ref. 103)

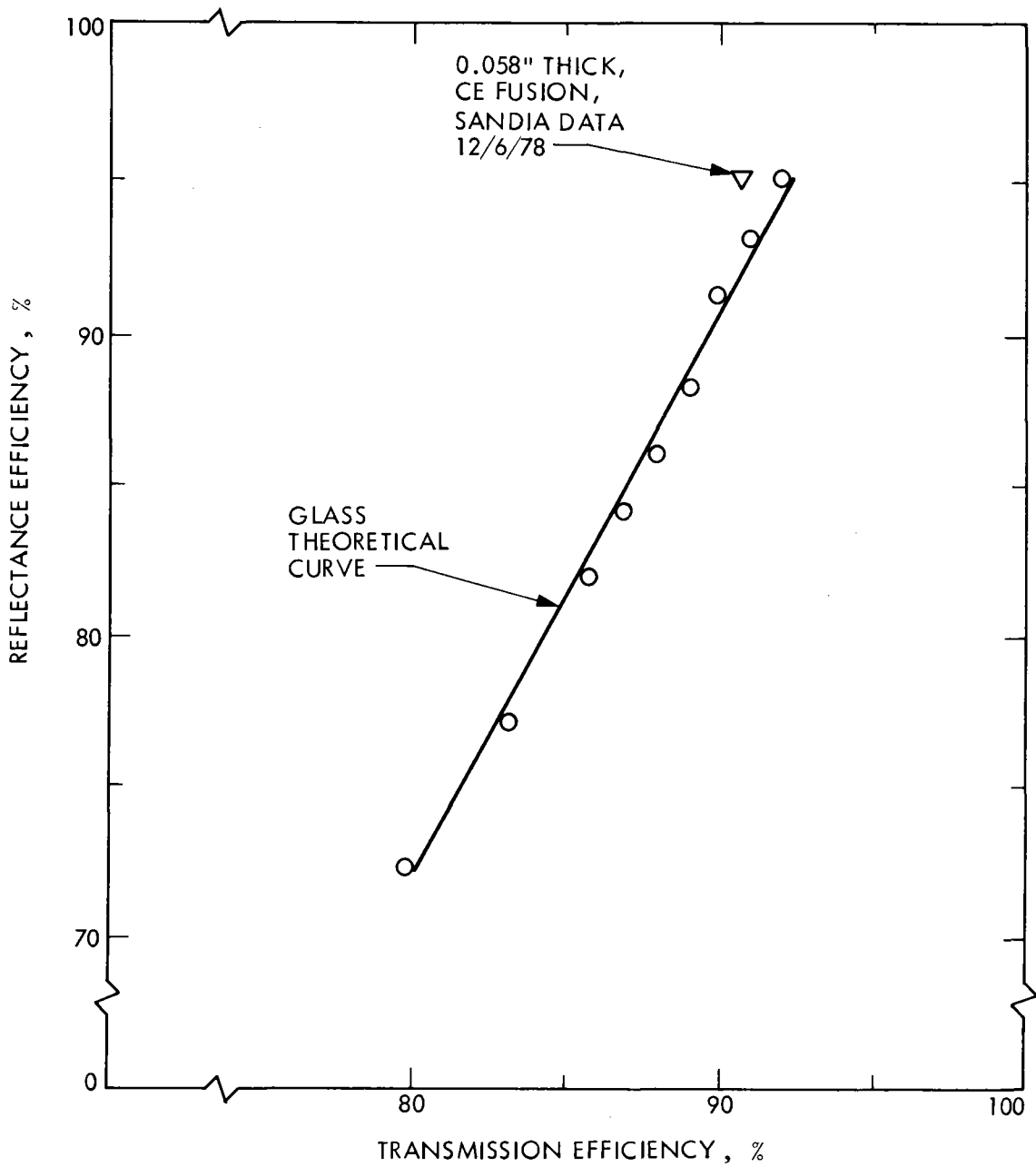


Figure 26. Glass Transmission and Reflection Efficiencies (Ref. 103)

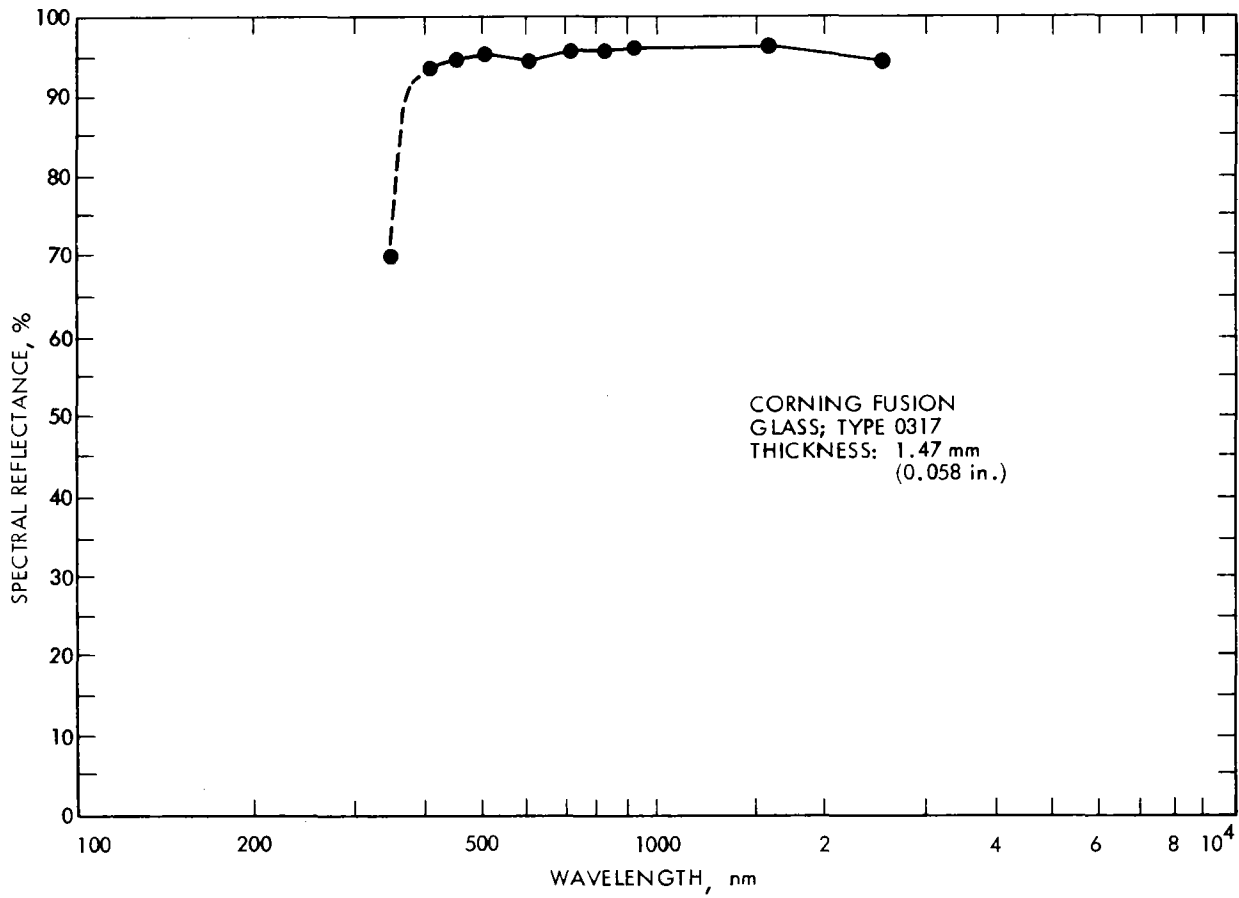


Figure 27. Reflectivity vs. Wavelength for Fusion Low Iron Glass (Ref. 111)

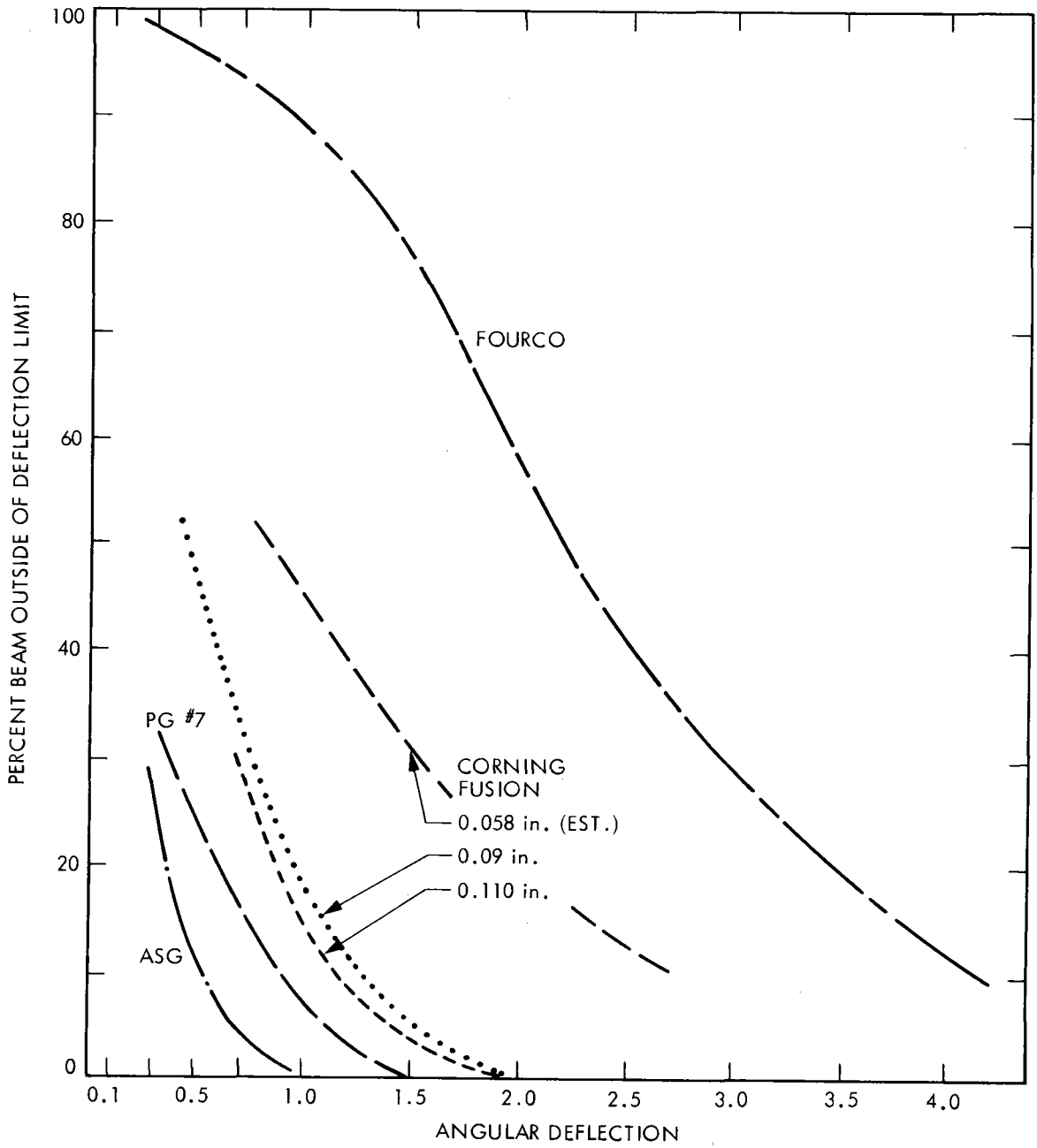


Figure 28. Percent Beam vs. Angular Deflection (Ref. 96)

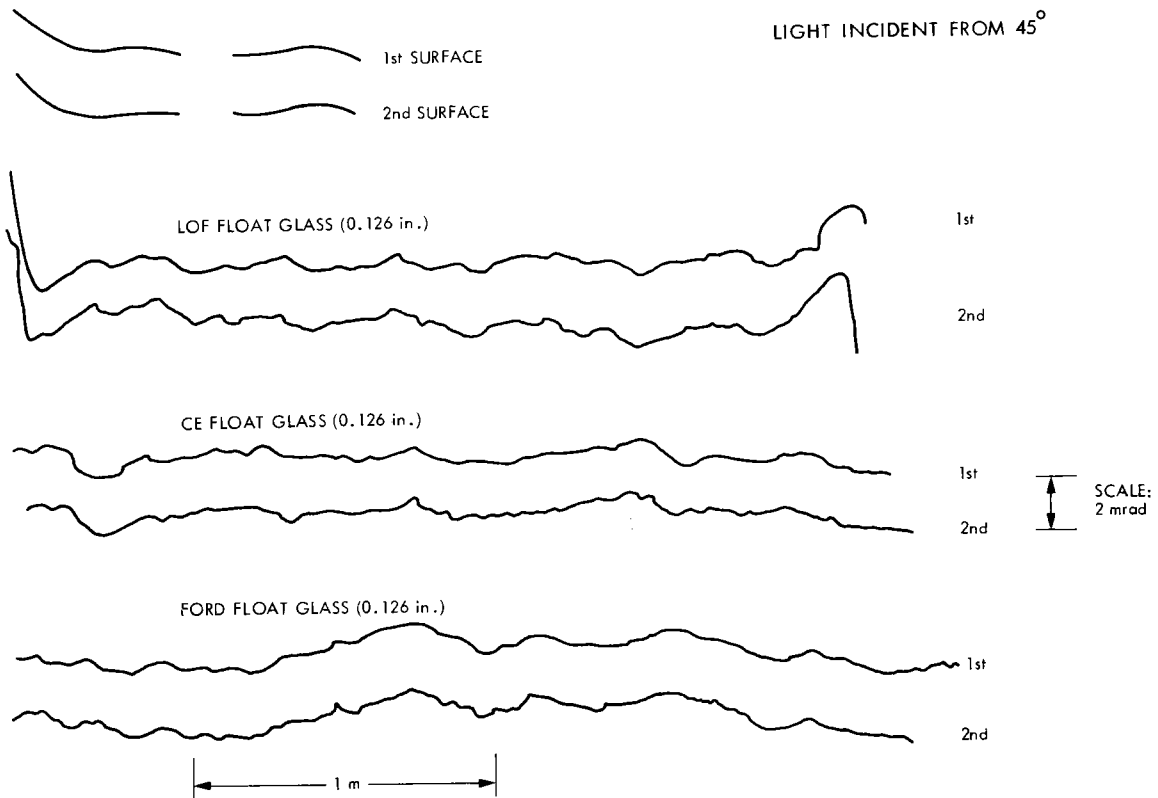


Figure 29. Angular Deviation From Specular Direction for First and Second Surface Reflections (Ref. 96)

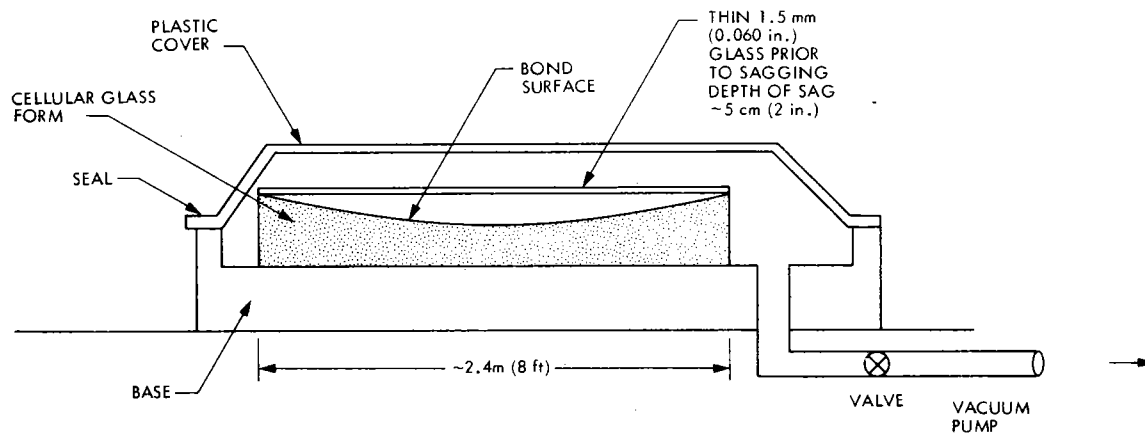


Figure 30. Schematic of Proposed 3D Vacuum Forming Technique

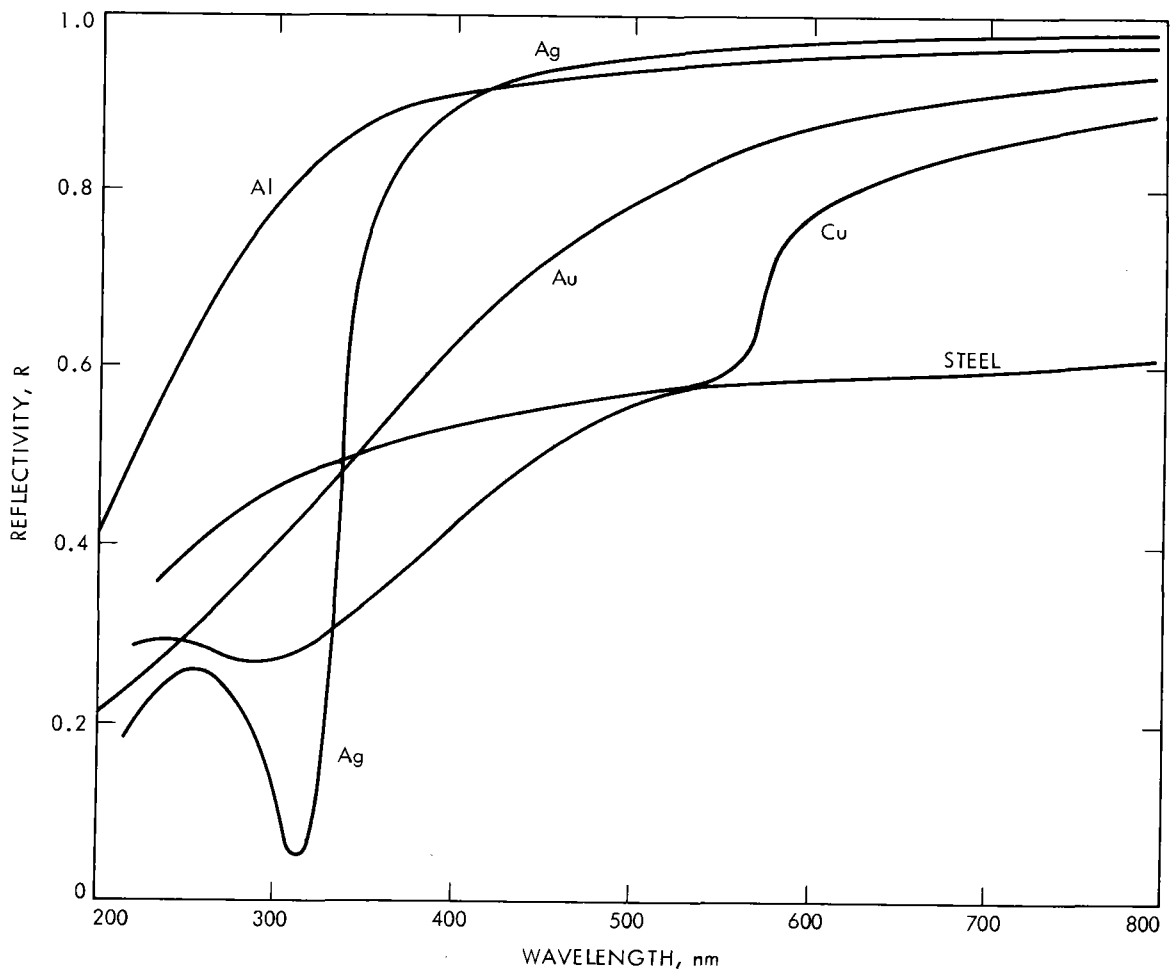


Figure 31. Reflectivity of Some Common Metals for Normal Incidence as a Function of Wavelength (Ref. 104)

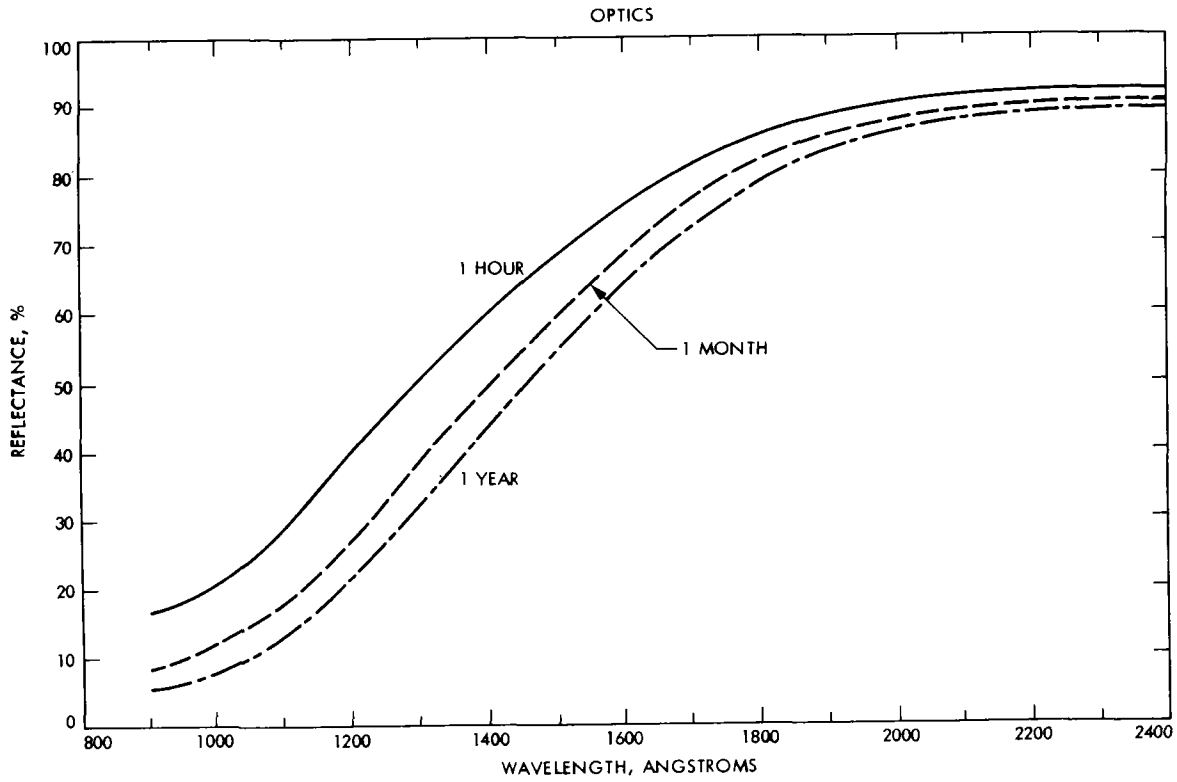


Figure 32. Degradation of Reflectance on Clean Aluminum Mirrors vs. Wavelength (Ref. 53)

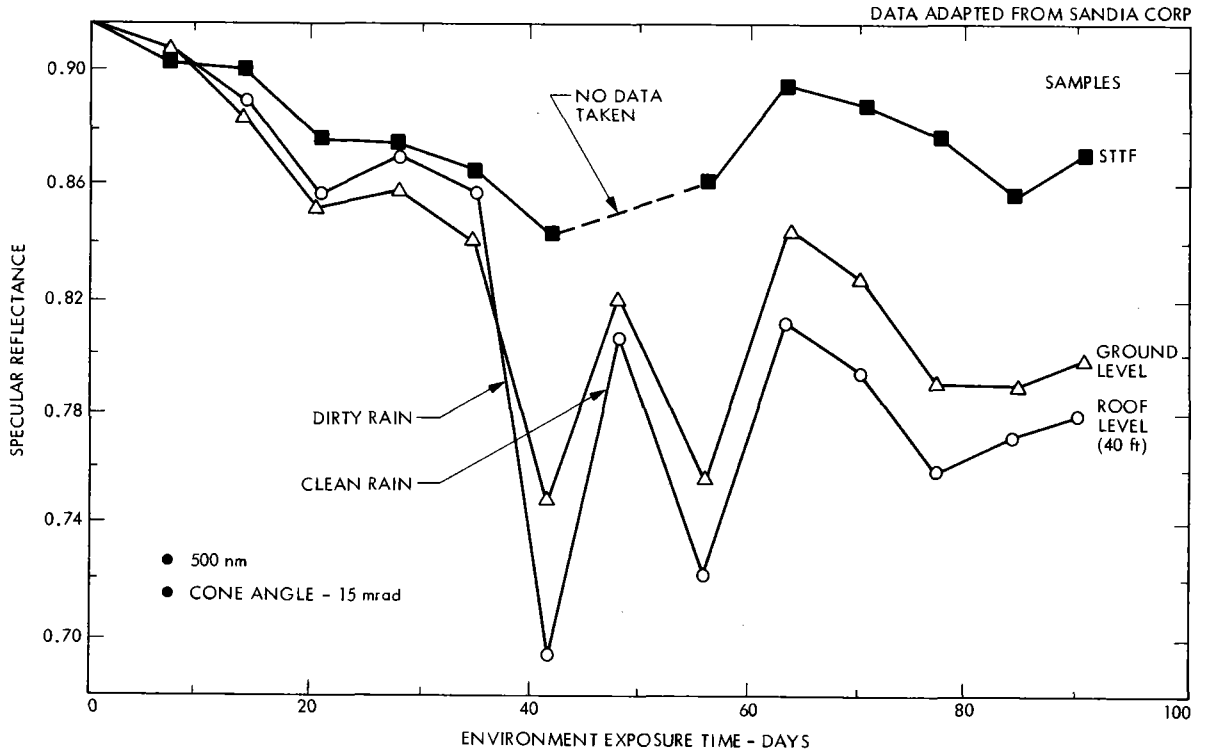


Figure 33. Glass Mirrors: Specular Reflectance vs. Environmental Exposure Time Location - Albuquerque, NM - 1978

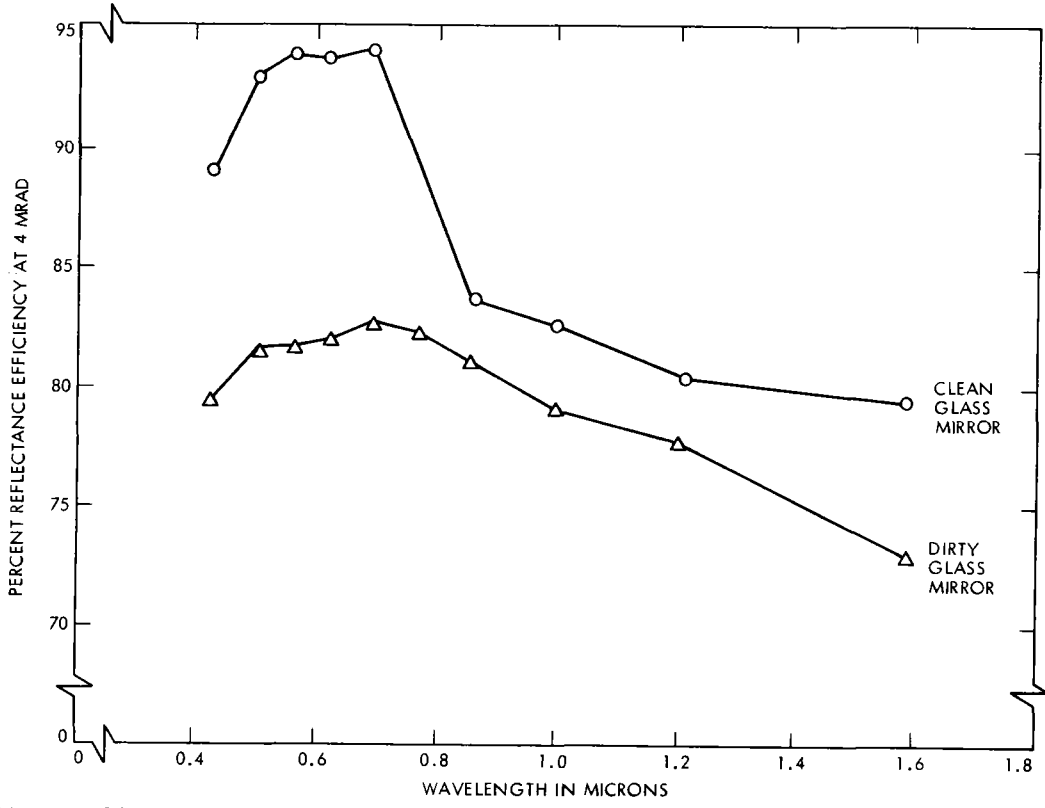


Figure 34. Percent Reflection Efficiency vs. Wavelength (Ref. 103)

Visual Appearance

Visual evaluations were made under artificial daylight (MacBeth Examolite type TC-440) and with an intense concentrated beam light source (Burton Lamp). The specimens were examined in both reflected and transmitted light and were evaluated as follows:

- A. (excellent) - No spots and/or haze are visible when examined with the concentrated beam light source less than 6 in. from the specimen.
- B. (good) - A few spots and/or a slight haze are visible only with lighting conditions as for A.
- C. (fair) - Many spots and/or much haze are visible only with lighting conditions as for A.
- D. (poor) - Some spots and/or haze are visible without the concentrated beam light source, but with artificial daylight.
- E. (very poor) - An excessive accumulation of weathering products is readily visible with artificial daylight.

Any appearance degradation which could not be removed by scrubbing with a cloth and water is considered to be permanent damage.

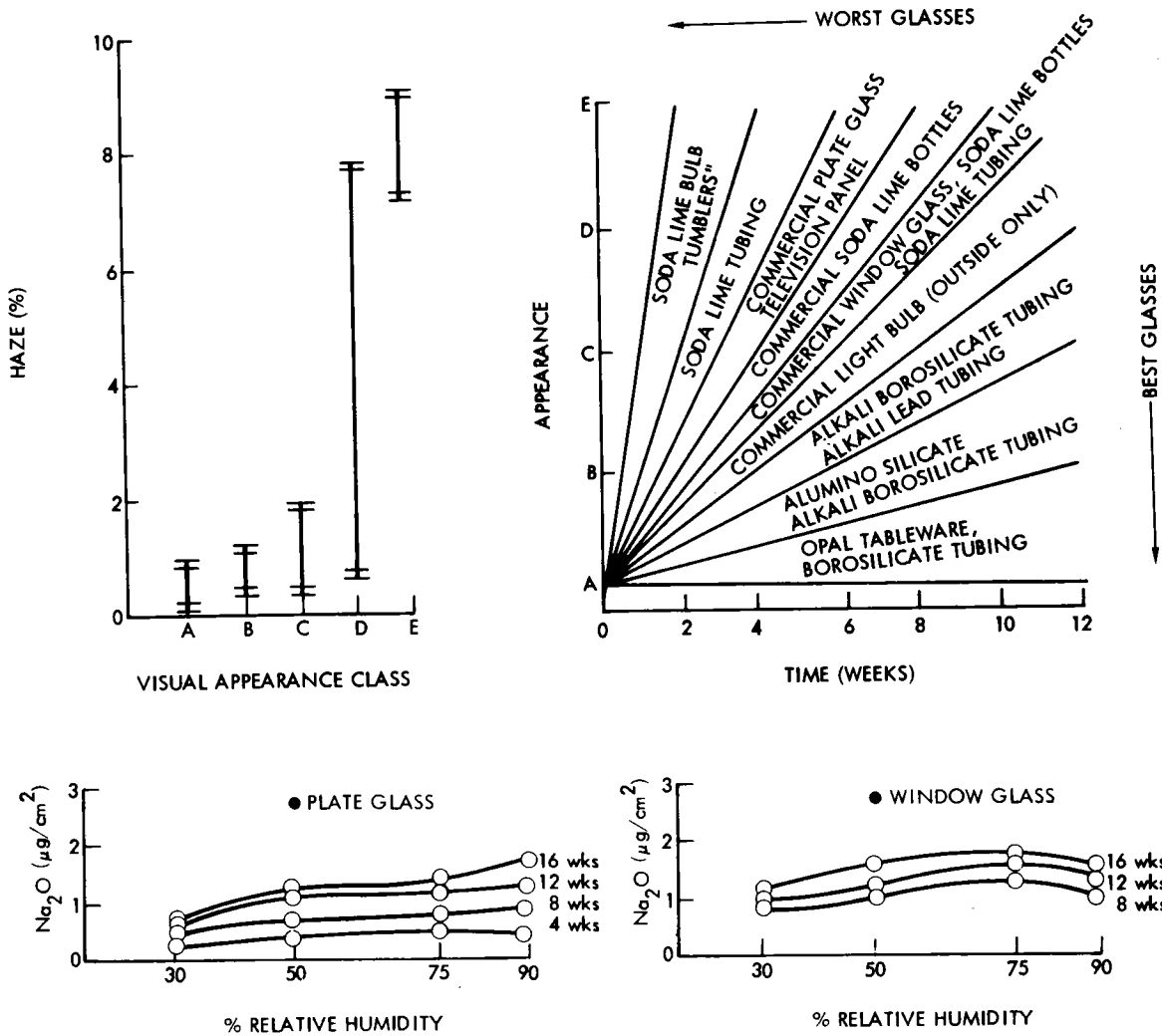


Figure 35. Cleaning Effects (Ref. 113)

Table 1. Analysis and Properties of Representative Glasses (Ref. 39)

| Type of Glass | Analysis, Percent by Weight | | | | | Softening Temp. °C | Coefficient of Expansion (°C) ⁻¹ × 10 ⁻⁷ |
|---------------------------|-----------------------------|-----------|--------------------------------|-------------------------------|-----|--------------------|--|
| | SiO ₂ | Modifiers | Al ₂ O ₃ | B ₂ O ₃ | PbO | | |
| Fused Silica | 99.9 | ----- | ----- | ----- | --- | 1667 | 5.5 |
| 96 percent Silica (Vycor) | 96.0 | ----- | ----- | 4.0 | --- | 1500 | 8.0 |
| Boro-silicate (Pyrex) | 80.5 | 4.2 | 2.2 | 12.9 | --- | 820 | 32.0 |
| Alumino-silicate | 57.7 | 9.5 | 25.3 | 7.4 | --- | 915 | 42.0 |
| Soda-lime Silica | 73.6 | 25.4 | 1.0 | ----- | --- | 696 | 92.0 |
| Lead-alkali | 54.0 | 11.0 | ----- | ----- | 35 | 630 | 89.0 |

Table 2. Glass Physical Properties

| Type of Glass | Specific Gravity g/cm ³ (lbs/ft ³) | Young's Modulus 10 ³ kg/mm ² (10 ⁶ psi) | Thermal Expansion* cm/cm°C (in/in°F) | Refractive Index | Poisson's Ratio |
|------------------|---|--|--|------------------|-----------------|
| Soda Lime | 2.47 (154) | 7.1 (10.2) | 93.6 × 10 ⁻⁷ (52 × 10 ⁻⁷) | 1.512 | 0.22-0.24 |
| Alumino-silicate | 2.52-2.64 (154.6-157.2) | 8.8-8.9 (12.5-12.7) | 42.1-46.1 × 10 ⁻⁷ (23.4-25.6 × 10 ⁻⁷) | 1.53-1.547 | 0.24-0.25 |
| Boro-silicate | 2.13-2.48 (132.8-154.6) | 5.0-6.9 (7.1-9.8) | 32-51.5 × 10 ⁻⁷ (17.8-28.6 × 10 ⁻⁷) | 1.473 | 0.2-0.23 |
| 96% Fused Silica | 2.18 (135.9) | 6.9 (9.8) | 7.6-8 × 10 ⁻⁷ (4.2-4.4 × 10 ⁻⁷) | 1.458 | 0.19 |
| Fused Silica | 2.2 (137.2) | 7.4 (10.5) | 5.6 × 10 ⁻⁷ (3.1 × 10 ⁻⁷) | 1.459 | 0.16 |

* Over the range 0 to 300°C or -18 to 572°F

Source: Corning Glass Works

Table 3. Modulus of Rupture Design Values (Adapted from Ref. 9)

| Nominal Thickness -cm (in) | Approximate Variation -cm (in) | | Flat Plates (Regular Temper) -N/m ² (psi) | Curved Plates (Regular Temper) -N/m ² (psi) |
|----------------------------------|--------------------------------------|------------------|---|---|
| | Max. | Min. | | |
| 0.3175 (0.125) | 0.287 (0.133) | 0.28 (0.110) | 96.5M (14,000) | 89.6M (13,000) |
| 0.476 (0.1875) | 0.056 (0.220) | 0.475 (0.187) | 137.8M (20,000) | 117.2M (17,000) |
| 0.635 (0.25) | 0.66 (0.260) | 0.559 (0.220) | 172.35M (25,000) | 137.8M (20,000) |

Table 4. Temperature Characteristics of Glasses

| Glasses | T ^{°C} | Softening Point (°C) | Thermal Expansion (Linear) Parts/°C |
|-------------------|-----------------|----------------------|-------------------------------------|
| Iron-sealing | 24 | 484 | 132x10 ⁻⁷ |
| Soda-borosilicate | 25 | 693 | 50x10 ⁻⁷ |
| Fused quartz | 25 | 1667 | 5.7x10 ⁻⁷ |

Table 5. Coefficients of Thermal Expansion (Ref. 118)

| Type of Glass | Temperature Interval | Coefficient ($\times 10^{-4}$) | Temperature Interval | Coefficient ($\times 10^{-4}$) |
|--------------------------|----------------------|----------------------------------|----------------------|----------------------------------|
| 1. Barium flint | 22-494 | 0.088 | 519-550 | 0.331 |
| 2. Plate glass | 20-508 | 0.108 | 540-560 | 0.401 |
| 3. Light crown | 24-422 | 0.104 | 494-507 | 0.548 |
| 4. Borosilicate crown | 22-498 | 0.090 | 539-562 | 0.393 |
| 5. Medium flint | 23-402 | 0.097 | 452-478 | 0.396 |
| 6. Commercial glass | 23-445 | 0.107 | 510-534 | 0.309 |
| 7. Pyrex | 21-471 | 0.036 | 552-571 | 0.151 |
| 8. Schott-Genosser flask | 19-414 | 0.056 | 540-562 | 0.404 |
| 9. Soda tubing | 21-372 | 0.120 | 506-525 | 0.234 |

Table 6. Refraction Indices of Glass - Relative to Air (Ref. 118)

| Type of Glass | 0.361 | 0.434 | Wavelength in Microns | | | | 1.20 | 2.00 |
|-------------------------|-------|--------|-----------------------|-------|-------|-------|-------|-------|
| | | | 0.486 | 0.589 | 0.656 | 0.768 | | |
| Zinc Crown | 1.539 | 1.5281 | 1.523 | 15.17 | 1.514 | 1.511 | 1.505 | 1.497 |
| Higher Dispersion Crown | 1.546 | 1.533 | 1.527 | 1.520 | 1.517 | 1.514 | 1.507 | 1.497 |
| Light Flint | 1.614 | 1.594 | 1.585 | 1.575 | 1.571 | 1.567 | 1.559 | 1.549 |
| Heavy Flint | 1.705 | 1.675 | 1.664 | 1.650 | 1.644 | 1.638 | 1.628 | 1.617 |
| Heaviest Flint | --- | 1.945 | 1.919 | 1.890 | 1.879 | 1.867 | 1.848 | 1.832 |

Table 7. Refractive Indices vs Temperature for Optical Grade Glass, Corning No. 7913 (Vycor) (Ref. 116)

| λ , μm | n, 28°C | n, 526°C | dn/dT ($10^{-6}/\text{°C}$) | n, 826°C | dn/dT ($10^{-6}/\text{°C}$) |
|---------------------------|---------|----------|----------------------------------|----------|----------------------------------|
| 0.26520 | 1.49988 | 1.50799 | +16.3 | 1.51438 | +18.2 |
| 0.28936 | 1.49074 | 1.49831 | +15.2 | 1.50418 | +16.8 |
| 0.29673 | 1.48851 | 1.49587 | +14.8 | 1.50164 | +16.5 |
| 0.30215 | 1.48694 | 1.49423 | +14.6 | 1.49990 | +16.2 |
| 0.3130 | 1.48416 | 1.49121 | +14.2 | 1.49679 | +15.8 |
| 0.33415 | 1.47949 | 1.48622 | +13.5 | 1.49158 | +15.2 |
| 0.36502 | 1.47415 | 1.48065 | +13.1 | 1.48570 | +14.5 |
| 0.40466 | 1.46925 | 1.47547 | +12.5 | 1.48027 | +13.8 |
| 0.43584 | 1.46628 | 1.47234 | +12.2 | 1.47708 | +13.5 |
| 0.54607 | 1.45960 | 1.46544 | +11.7 | 1.46992 | +12.9 |
| 0.5780 | 1.45831 | 1.46407 | +11.6 | 1.46849 | +12.8 |
| 1.01398 | 1.44968 | 1.45526 | +11.2 | 1.45924 | +12.0 |
| 1.12866 | 1.44831 | 1.45373 | +10.9 | 1.45779 | +11.9 |
| 1.254 | 1.44677 | 1.45222 | +10.9 | 1.45627 | +11.9 |
| 1.36728 | 1.44554 | 1.45095 | +10.9 | 1.45504 | +11.9 |
| 1.470 | 1.44422 | 1.44965 | +10.9 | 1.45370 | +11.9 |
| 1.52952 | 1.44356 | 1.44896 | +10.8 | 1.45306 | +11.9 |
| 1.660 | 1.44206 | 1.44750 | +11.0 | 1.45157 | +11.9 |
| 1.701 | 1.44137 | 1.44677 | +10.8 | 1.45088 | +11.9 |
| 1.981 | 1.43750 | 1.44291 | +10.9 | 1.44702 | +11.9 |
| 2.262 | 1.43298 | 1.43839 | +10.9 | 1.44258 | +12.0 |
| 2.553 | 1.42825 | 1.43373 | +11.0 | 1.43824 | +12.5 |

Table 8. Temperature Coefficients of Refractive Index (Ref. 36)

| Glass | dn/dT $\times 10^6$ | $\alpha \times 10^6$ | dF/F dT $\times 10^6$ | dP/P dT $\times 10^6$ |
|----------------------------|------------------------|----------------------|--------------------------|--------------------------|
| Silica (n - 1 = 0.460) | +9.3 | 0.6 | -19.6 | +20.8 |
| 513-637 Borosilicate crown | +1.4 | 7.8 | +5.1 | +10.5 |
| 517-602 Silicate crown | -1.1 | 9.2 | +11.3 | +7.1 |
| 571-430 Light flint | +2.5 | (9.1)* | +4.7 | +13.5 |
| 755-275 Dense flint | +7.8 | 8.0 | -2.3 | +18.3 |
| 573-574 Light barium crown | +0.4 | (8.3)* | +7.6 | +9.0 |
| 610-574 Dense barium crown | +4.1 | (7.1)* | +0.4 | +13.8 |

*Probable values.

Table 9. ASTM Tests Pertaining to Glass
(Ref. 24 and 58)

1. Test for Annealing Point and Strain Point of Glass by Beam Bending, C 598, Vol. 17.
2. Definition of Terms Relating to Glass and Glass Products, C 162, Vol. 17.
3. Standard Reference Materials for Glass and Glass Products, Vol. 17.
4. Recommended Practices for Glass Stress Optical Coefficient, C 770, Vol. 17.
5. Test for Hydrophobic Contamination on Glass by Water Condensation, C 812, Vol. 17.
6. Test for the Softening Point of Glass, C 338, Vol. 17.
7. Test for Analyzing Stress in Glass, F 218, Vol. 17, 43.
8. Test for Young's Modulus, Shear Modulus and Poisson's Ratio for Glass and Glass-Ceramics by Resonance, C 623, Vol. 17.
9. Test for Linear Expansion . . . E 228, Vol. 10, 17, 41, 44.
10. Hydrophobic Contamination Test on Glass by Contact Angle, C 813, Vol. 17.

Table 10. Summary of Reflective Surfaces for Solar Concentrator

| No. | Producer/Supplier | Material Type | Glass Thickness mm (in) | Hemispherical Solar Reflectance | Glass Thickness mm (in) | Solar Transmittance | Reflectance at 500 nm* | | | | Coef. Thermal Expansion -cm/cm°C | Cost, † \$/Mft ² | Remarks |
|-----|------------------------|-------------------------|----------------------------|------------------------------------|----------------------------|---------------------|------------------------|---------------------------|----------------|---------------------------|-------------------------------------|--------------------------------|--|
| | | | | | | | R ₁ | σ ₁ (m rad) | R ₂ | σ ₂ (m rad) | | | |
| 1 | Alcoa | Alzak | 0 | 0.85 | 0 | 0 | 0.56 (505) | 0.42 | 0.33 | 10.1 | NA | NA | Aluminum |
| 2 | Alcoa | S460666 | 0 | 0.32 | 0 | 0 | -- | -- | -- | -- | NA | NA | Aluminum |
| 3 | Alcoa | S460667 | 0 | 0.92 | 0 | 0 | -- | -- | -- | -- | NA | NA | Aluminum |
| 4 | ISC | 90-10 | 0 | 0.90 | 0 | 0 | 0.86 | -- | -- | -- | -- | 1.50 | Silver plated brass |
| 5 | ISC | 80-20 | 0 | 0.88 | 0 | 0 | 0.80 | -- | -- | -- | -- | 1.50 | Requires overcoat Cu-Zn |
| 6 | ISC | 70-30 | 0 | 0.91 | 0 | 0 | 0.81 | -- | -- | -- | -- | 1.50 | 80% Cu, 20% Zn |
| 7 | 3M | Scotchcal 5400 | 0 | 0.85 | 0 | 0 | 0.96 | 1.9 | -- | -- | -- | 0.5 (E) | Estimated cost |
| 8 | Corning | Code 7806 (fusion) | -- | -- | 1.14 (0.045) | 0.88 | -- | -- | -- | -- | 70 (-7) (0-300°C) | 1.40 | >10 Mft ² , \$0.45 |
| 9 | Corning | Code 0317 (fusion) | -- | -- | 2.29 (0.090) | 0.910 | -- | -- | -- | -- | 88 (-7) (0-300°C) | 0.65-0.80 1.10-1.80 | Without metallization |
| 10 | Corning | Code 0317 (fusion) | -- | -- | 1.52 (0.060) | 0.909 | -- | -- | -- | -- | 88 (-7) (0-300°C) | | With metallization |
| 11 | Corning | Code 0317 (fusion) | -- | -- | 2.8 (0.110) | 0.903 | -- | -- | -- | -- | 88 (-7) (0-300°C) | | |
| 12 | Schott B270 | B270 (Rolled) | -- | -- | 3 (0.120) | 0.913 | -- | -- | -- | -- | NA | 0.5-0.8 | Without metallization |
| 13 | PPG Works | #6 (Float) | -- | -- | 3.17 (0.125) | 0.881 | -- | -- | -- | -- | 86 (-7) (25-300°C) | 2.15 | >10 ⁷ ft ² , \$0.60-0.65 |
| 14 | ASG | (Float) | -- | -- | 3.17 (0.125) | 0.847 | -- | -- | -- | -- | 85 (-7) (0-300°C) | 0.30 | |
| 15 | Sheldahl | Aluminized Teflon | NA | 0.87 | -- | -- | 0.80 | 1.3 | 0.07 | 30.9 | NA | NA | |
| 16 | Kingston Ind. | Kingflux (Al) | 0 | <0.85 | 0 | 0 | 0.65 (498) | 0.37 | 0.23 | 16.1 | NA | 2.00 | Similar to Alzak; JPL |
| 17 | Corning | Microsheet Al | -- | 0.95 | 0 | 0 | 0.77 (550) | 1.1 | 0.18 | 6.2 | NA | NA | measurements, 85.4% |
| 18 | Carolina Mirror Co. | 2nd Surface Ag Glass | -- | 0.83 | 0 | 0 | 0.92 | 0.15 | -- | -- | NA | NA | small quantities |
| 19 | Payne Co. | Microglass | 0.15 (0.006) | 0.94 | -- | -- | -- | -- | -- | -- | NA | NA | |
| 20 | Payne Co. | Microglass | 0.30 (0.012) | 0.93 | -- | -- | -- | -- | -- | -- | NA | NA | |

* Measurements at other wavelengths are shown in parentheses. Data from R. B. Pettit.

† These costs are preliminary and are being updated.

(Continued next page)

Table 10 (Cont.). Summary of Reflective Surfaces for Solar Concentrator

| No. | Producer/ Supplier | Material Type | Glass Thickness mm (in) | Hemispherical Solar Reflectance | Glass Thickness mm (in) | Solar Transmittance | Reflectance at 500 nm* | | | | Coef. Thermal Expansion -cm/cm°C | Cost † \$/Hft ² | Remarks |
|-----|-----------------------|--|-------------------------------|------------------------------------|-------------------------------|------------------------|------------------------|---------------------------|----------------|---------------------------|--|-------------------------------|---|
| | | | | | | | R ₁ | σ ₁ (m rad) | R ₂ | σ ₂ (m rad) | | | |
| 21 | CE | Glass (float) (Soda lime) | -- -- | -- -- | 3.17 (0.125) | 0.838 | -- | -- | -- | -- | 85 (-7) (0-300°C) | 0.50 | |
| 22 | Ford | Glass (float) (soda lime) | -- | -- | 3.17 (0.125) | 0.844 | -- | -- | -- | -- | 85 (-7) (0-300°C) | 0.40 | |
| 23 | Fourco | Glass (float) (Soda lime) | -- | -- | 3.17 (0.125) | 0.891 | -- | -- | -- | -- | -- | NA | |
| 24 | Liberty Mirrors | Cr coated front surface glass | 3.17 (0.125) | 0.65 | -- | -- | -- | -- | -- | -- | 32-93(-7) (20-300°C) | NA | Special measurements |
| 25 | | Lead-sulfide front surface glass | 3.17 (0.125) | 0.25 | -- | -- | -- | -- | -- | -- | 32(-7) (20-300°C) | NA | Auto side mirrors applications only |
| 26 | Schott-Jena | Tempax (sheet) | -- | -- | -- | -- | -- | -- | -- | -- | 32 (-7) (20-300°C) | NA | |
| 27 | Corning | Code 0317 (fusion low Fe) | 1.47 (0.058) | 0.95 ±1 | -- | -- | -- | -- | -- | -- | 88 (-7) (0-300°C) | + | Sandia data + see item #9 vacuum deposited silver |
| 28 | Corning | Code 0317 (fusion low Fe) | 1.47 (0.058) | 0.94 ±1 | -- | -- | -- | -- | -- | -- | 88 (-7) (0-300°C) | + | Sandia data + see item #9 chemically deposited silver |
| 29 | Corning | Code 0317 (fusion low Fe) | 1.47 (0.058) | 0.926 | -- | -- | -- | -- | -- | -- | 88 (-7) (0-300°C) | + | JPL measurements old glass. 12/1/78 |
| 30 | 3M | Metallized Polyester | 0.07 (0.0028) | 0.86 | -- | -- | 0.86 (E) | 1.9 (E) | -- | -- | -- | 0.50 | Measurements @ AM2 20% degradation/7 yrs. |
| 31 | Flabeg Corp. | Crown Glass (float) | 3.17 (0.125) | TBD | -- | -- | -- | -- | -- | -- | 70 (-7) (0-300°C) | -- | Resin and/or tyilar reverse side sealant |
| 32 | Corning | 7806 (modified) (fusion) | 0.050(E) | 0.95 (E) | -- | -- | -- | -- | -- | -- | 70 (-7) (0-300°C) | -- | |

* Measurements at other wavelengths are shown in parentheses. Data from R. B. Pettit.

† These costs are preliminary and are being updated.

E Estimated

Table 11. Glass Survey Summary (Ref. 96)

| Manufacturer | Process | Composition | Thickness | | Solar Transmittance | | Optical Flatness % Power Outside | | Approximate Cost | |
|--------------|-----------------------|-------------------|-----------|----------|---------------------|----------|-------------------------------------|-----------|------------------------|--------------------------|
| | | | Tested | Possible | Measured | Possible | 1 mrad | 2 mrad | 1M, ft ² | >10M, ft ² |
| | | | | | | | | | | |
| ASG | Lo-Iron Float | Soda Lime | 0.125 | | 0.847 | | | | 0.30 | |
| | Sunadex-Twin Ground | Soda Lime | | >4mm | | >0.91 | 0* | 0* | 1.30 | |
| CE | Float | Soda Lime | 0.125 | | 0.838 | | 18 | 1 | 0.50 | |
| | Heliolite-Twin Ground | Soda Lime | | | | 0.91 | 0* | 0* | 1.30 | |
| Corning | 0317 Fusion | Aluminosilicate | 0.110 | >0.020 | 0.903 | | 15 | 0 | 0.65 - 0.80 | |
| | 0317 Fusion | Aluminosilicate | 0.090 | | 0.910 | | 19 | 0 | 0.65 - 0.80 | |
| | 7806 Fusion | Lime Borosilicate | 0.045 | | 0.876** | >0.91 | | | 1.40 | 0.45 |
| Fourco | Rolled | Soda Lime | 0.125 | | 0.891 | | 90 | 57 | | |
| Ford | Float | Soda Lime | 0.125 | >0.105 | 0.844 | >0.88 | 5 | 0 | 0.40 | |
| Guardian | Float | Soda Lime | | >0.085 | | >0.88 | | | 1.00 | |
| LOF | Float | Soda Lime | 0.125 | | 0.831 | | 7 | 0 | | |
| PPG | Mid-Iron Float | Soda Lime | 0.125 | | 0.866 | | 8 | 0 | | |
| | Lo-Iron Float | Soda Lime | 0.125 | >0.060 | 0.881 | >0.89 | 20 | 4 | 2.15 | 0.60 - 0.65 |

* Based on Measurements of ASG Tinted Twin Ground Glass

** Measured by R. Pettit, SLA, Total Hemispherical Solar Transmittance

Table 12. Reflectance Efficiency of Glass Mirrors at AM2 (Ref. 103)

| Glass (Wt Percent Iron Oxide) | Mirror Manufacturing and Silvering Process | Thickness in mm (In.) | Specimen No. | Over Solar Spectrum using Beckman 125 mrad | At Wavelength of 560 nm using Specular Photometer | | | Over Solar Spectrum using Specular Spectro Photometer | | |
|--|---|-----------------------------|---------------------------------|---|---|------------------------------|------------------------------|---|--------------|--------------|
| | | | | | 16 mrad | 8 mrad | 4 mrad | 16 mrad | 8 mrad | 4 mrad |
| ASG Ind Lustra Sheet (0.05 - 0.06) | Buchmin Industries Mirrorlab Process | 0.7 (0.028) | 134.1 | 96.0 | | | | 94.0 | 94.0 | |
| | | 2.4 (0.093) | 110.1 110.2 | 94.0 94.0 | | | | 93.0 92.0 | 92.0 92.0 | 86.0 88.0 |
| | Gardner Mirror and Process | 3.2 (0.125) | 65.3 | 91.4 | 95.5 | 95.0 | 93.5 | | | |
| | | | 65.13 65.15 65.16 67.3 | 93.5 93.5 94.3 91.9 | 94.5 94.5 95.0 95.5 | 94.0 94.5 95.0 94.5 | 93.5 93.5 94.0 94.0 | 92.0 | 93.0 | 87.0 |
| Binswanger Mirror Co. Mirrorlab Process | 4.7 (0.187) | 114.1 | 88.0 | | | | 86.0 | 86.0 | | |
| Fourco Sheet (0.06) | Tyre Bros. Two Part London Laboratory Process | 2.3 (0.090) | 32.3 | 93.1 | 95.0 | 94.0 | 90.0 | | | |
| | | | 50.2 50.3 50.4 50.5 | 90.4 90.3 90.7 90.3 | 96.0 97.0 96.5 95.0 | 96.0 96.0 96.5 95.0 | 94.0 96.0 96.0 92.0 | | | |
| | | 6.4 (0.250) | 33.3 | 86.8 | 89.0 | 89.0 | 88.0 | | | |
| | | | 51.2 51.3 51.4 51.5 | 83.9 84.4 84.5 84.3 | 94.0 93.0 93.0 91.0 | 94.0 92.0 93.0 90.0 | 88.0 88.0 91.0 88.5 | | | |
| PPG Industries Float | (0.05) | 3.2 (0.125) | 1 | 92.0 | 95.0 | 95.0 | 95.0 | 90.0 | 89.0 | 81.0 |
| | | | 2 | 92.0 | 95.0 | 95.0 | 89.0 | 88.0 | 81.0 | |
| | | | 3 | 92.0 | 95.0 | 96.0 | 90.0 | 89.0 | 84.0 | |
| | (0.07) | 2.4 (0.093) | 111.1 | 88.0 | | 95.0 | | 87.0 | 86.0 | 82.0 |
| | | | 111.2 | 88.0 | | | 87.0 | 86.0 | 81.0 | |
| | (0.10) | 3.2 (0.125) | 86.1 | 83.5 | | | | 83.3 | 82.8 | 77.1 |

(Continued next page)

Table 12 (Cont.). Reflectance Efficiency of Glass Mirrors at AM2

| Glass (Wt Percent Iron Oxide) | Mirror Manufacturer and Silvering Process | Thickness in mm (In.) | Specimen No. | Over Solar Spectrum using Beckman 125 mrad | At Wavelength of 560 nm using Specular Photometer | | | Over Solar Spectrum using Specular Spectro Photometer | | | |
|--|---|-----------------------------|-----------------|---|---|--------------|--------------|---|-----------|-----------|------|
| | | | | | 16 mrad | 8 mrad | 4 mrad | 16 mrad | 8 mrad | 4 mrad | |
| Ford Motor Float (0.09 - 0.10) | Buchmin Industries Mirrorlab Process | 2.4 (0.093) | 52.2 | 85.7 | 96.5 | 96.0 | 96.0 | | | | |
| | | | 52.3 | 85.7 | 94.5 | 94.5 | 94.0 | | | | |
| | | | 52.4 | 85.9 | 94.0 | 93.5 | 93.0 | | | | |
| 52.5 | 86.0 | | 96.0 | 95.0 | 95.0 | | | | | | |
| 52.6 | 86.1 | | 93.0 | 93.0 | 93.0 | | | | | | |
| 3.2 (0.125) | 65.x 65.22 | 83.3 83.2 | 91.0 91.0 | 91.0 91.0 | 86.0 88.0 | 81.9 81.0 | 81.9 81.0 | 78.0 76.0 | | | |
| 6.4 (0.250) | 6.001 6.002 | 70.7 70.7 | 89.0 89.0 | 88.5 89.5 | 88.5 88.5 | | | | | | |
| Guardian Ind Float (0.10) | Buchmin Ind Mirrorlab Process | 3.2 (0.125) | 66.3 | 82.5 | 93.5 | 92.5 | 92.0 | | | | |
| ASG Industries Float (0.10 - 0.13) | Binswanger Mirror Co. Mirrorlab Process | 2.4 (0.093) | | | | | | | | | |
| | | | 3.2 (0.125) | Lot No. 2 | 83.7 | 92.0 | 91.0 | 90.5 | 81.6 | 82.1 | 78.5 |
| | | | | B-1 | 84.2 | 92.0 | 92.0 | 86.0 | 83.1 | 83.1 | 78.8 |
| | B-2 | 83.7 | | 92.0 | 92.0 | 89.5 | 82.3 | 82.3 | 75.9 | | |
| B-3 | 84.8 | 92.5 | | 92.0 | 87.5 | 84.8 | 84.8 | 80.9 | | | |
| 4.8 (0.188) | Lot No. 1 | 77.9 | 88.0 | 88.0 | 81.5 | 76.0 | 76.0 | 71.8 | | | |
| Buchmin Ind. Mirrorlab Process | 6.4 (0.250) | 64.22 | 71.1 | 88.0 | 88.0 | 88.0 | 71.0 | 71.0 | 66.0 | | |
| LOF Float (0.12) | Tyre Bros. 2-Part London Lab Process | 2.5 (0.100) | 29.3 | 82.4 | 93.0 | 92.0 | 92.0 | | | | |
| | | | 30.1 | 64.7 | 81.0 | 79.0 | 75.0 | | | | |

Table 13. Solar Transmission Efficiencies of Some Coatings
(Ref. 103)

| Supplier | Specimen Identification | Over Solar Spectrum using Beckman | At Wavelength of 550 nanometer using Specular Photometer | | |
|-------------------|--------------------------------------|-----------------------------------|--|--------|--------|
| | | 125 mrad | 16 mrad | 8 mrad | 4 mrad |
| De Soto Inc. | 1091-3-51 Fluorocarbon Acrylate | 99.9 | 100 | 99.5 | 99.0 |
| | 1091-6-51 Polyurethane | 99.9 | 99.5 | 98.5 | 98.5 |
| | 1091-10-51 Thermosetting Silicone | 99.7 | 98.5 | 98.5 | 98.0 |
| Rohm and Haas Co. | R&H No. 5 | 99.6 | 99.5 | 100 | 99.5 |
| | R&H No. 7 | 99.7 | 99.0 | 99.0 | 98.0 |
| | R&H No. 12 | 99.3 | 99.0 | 98.5 | 97.0 |
| PPG Industries | PPG No. 2 | 99.5 | 100 | 99.5 | 99.5 |
| | PPG No. 5 | 99.5 | 99.0 | 98.5 | 99.0 |
| DuPont Co. | 500 S IMRON | 99.5 | 100 | 100 | 100 |
| | 326 L Acrylic | 98.5 | 99.5 | 99.5 | 99.0 |
| | RK-63654 | 99.3 | 99.0 | 99.0 | 97.0 |
| O'Brien Corp. | 874-C-200 | 99.5 | 100 | 99.0 | 99.5 |
| | 5-73C | 99.4 | 99.0 | 99.0 | 98.0 |

Table 14. Optical Properties of Metals (Ref. 53)
 Percent Normal-Incidence Reflectance of Freshly
 Evaporated Mirror Coatings of Aluminum, Silver,
 Gold, Copper, Rhodium, and Platinum, from the
 Ultraviolet to the Infrared*

| λ , μm | Al | Ag | Au | Cu | Rh | Pt |
|---------------------------|------|------|------|------|------|------|
| 0.220 | 91.5 | 28.0 | 27.5 | 40.4 | 57.8 | 40.5 |
| 0.240 | 91.9 | 29.5 | 31.6 | 39.0 | 63.2 | 46.9 |
| 0.260 | 92.2 | 29.2 | 35.6 | 35.5 | 67.7 | 51.5 |
| 0.280 | 92.3 | 25.2 | 37.8 | 33.0 | 70.7 | 54.9 |
| 0.300 | 92.3 | 17.6 | 37.7 | 33.6 | 73.4 | 57.6 |
| 0.315 | 92.4 | 5.5 | 37.3 | 35.5 | 75.0 | 59.4 |
| 0.320 | 92.4 | 8.9 | 37.1 | 36.3 | 75.5 | 60.0 |
| 0.340 | 92.5 | 72.9 | 36.1 | 38.5 | 76.9 | 62.0 |
| 0.360 | 92.5 | 88.2 | 36.3 | 41.5 | 78.0 | 63.4 |
| 0.380 | 92.5 | 92.8 | 37.8 | 44.5 | 78.1 | 64.9 |
| 0.400 | 92.4 | 95.6 | 38.7 | 47.5 | 77.4 | 66.3 |
| 0.450 | 92.2 | 97.1 | 38.7 | 55.2 | 76.0 | 69.1 |
| 0.500 | 91.8 | 97.9 | 47.7 | 60.0 | 76.6 | 71.4 |
| 0.550 | 91.5 | 98.3 | 81.7 | 66.9 | 78.2 | 73.4 |
| 0.600 | 91.1 | 98.6 | 91.9 | 93.3 | 79.7 | 75.2 |
| 0.650 | 90.5 | 98.8 | 95.5 | 96.6 | 81.1 | 76.4 |
| 0.700 | 89.7 | 98.9 | 97.0 | 97.5 | 82.0 | 77.2 |
| 0.750 | 88.6 | 99.1 | 97.4 | 97.9 | 82.6 | 77.9 |
| 0.800 | 86.7 | 99.2 | 98.0 | 98.1 | 83.1 | 78.5 |
| 0.850 | 86.7 | 99.2 | 98.2 | 98.3 | 83.4 | 79.5 |
| 0.900 | 89.1 | 99.3 | 98.4 | 98.4 | 83.6 | 80.5 |
| 0.950 | 92.4 | 99.3 | 98.5 | 98.4 | 83.9 | 80.6 |
| 1.0 | 94.0 | 99.4 | 98.6 | 98.5 | 84.2 | 80.7 |
| 1.5 | 97.4 | 99.4 | 99.0 | 98.5 | 87.7 | 81.8 |
| 2.0 | 97.8 | 99.4 | 99.1 | 98.6 | 91.4 | 81.8 |
| 3.0 | 98.0 | 99.4 | 99.3 | 98.6 | 95.0 | 90.6 |
| 4.0 | 98.2 | 99.4 | 99.4 | 98.7 | 95.8 | 93.7 |
| 5.0 | 98.4 | 99.5 | 99.4 | 98.7 | 96.4 | 94.9 |
| 6.0 | 98.5 | 99.5 | 99.4 | 98.7 | 96.8 | 95.6 |
| 7.0 | 98.6 | 99.5 | 99.4 | 98.7 | 97.0 | 95.9 |
| 8.0 | 98.7 | 99.5 | 99.4 | 98.8 | 97.2 | 96.0 |
| 9.0 | 98.7 | 99.5 | 99.4 | 98.8 | 97.4 | 96.1 |
| 10.0 | 98.7 | 99.5 | 99.4 | 98.9 | 97.6 | 96.2 |
| 15.0 | 98.9 | 99.6 | 99.4 | 99.0 | 98.1 | 96.5 |
| 20.0 | 99.0 | 99.6 | 99.4 | | | |
| 30.0 | 99.2 | 99.6 | 99.4 | | | |

*The reflectance of a good evaporated mirror coating is always higher than that of a polished or electroplated surface of the same material.

Table 15. Solar Efficiency of Silvering Processes* (Ref. 103)

| Silvering Process | Silver Thickness gm/m ² | Over Solar Spectrum using Beckman | At of 550 nm using Specular Photometer | | |
|--------------------|---------------------------------------|-----------------------------------|--|--------|--------|
| | | 125 mrad | 16 mrad | 8 mrad | 4 mrad |
| Peacock Laboratory | 1.35 (125 mg/ft ²) | 85.0 | 93.5 | 91.0 | 62.0 |
| London Laboratory | 0.86 (80 mg/ft ²) | 84.3 | 91.0 | 90.0 | 88.5 |
| | 0.84 (78 mg/ft ²) | 84.5 | 93.0 | 92.5 | 85.0 |
| Hilemn | 0.82 (76 mg/ft ²) | 84.1 | 92.5 | 91.5 | 79.0 |
| Mirrolab | 1.01 (94 mg/ft ²) | 85.3 | 92.5 | 90.0 | 59.5 |

* 6.4 mm (1/4 inch) thick Fourco sheet glass was used.

Table 16. Solar Efficiency versus Silver Thickness (Ref. 103)

| Type Glass and Thickness in millimeters (inches) | Specimen No. | Silver Deposition in grams per square meter | Beckman Spectro Photometer (Over Solar Spectrum) | Specular Photometer (550 nm) | | | Specular Spectrophotometer (Over Solar Spectrum) | | |
|--|--------------|---|--|------------------------------|--------|--------|--|--------|--------|
| | | | 125 mrad | 16 mrad | 8 mrad | 4 mrad | 16 mrad | 8 mrad | 4 mrad |
| Fourco Sheet Glass 6.4 mm (1/4-inch) | 58.4 | 0.65 (60 mg/ft ²) | 83.7 | 91.5 | 90.0 | 81.0 | | | |
| | 59.3 | 0.86 (80 mg/ft ²) | 84.1 | 91.5 | 91.0 | 89.0 | | | |
| | 60.1 | 1.01 (94 mg/ft ²) | 86.3 | 92.5 | 90.0 | 59.5 | | | |
| | 61.3 | 1.08 (100 mg/ft ²) | 83.1 | 90.0 | 88.0 | 82.0 | | | |
| | 62.1 | 1.29 (120 mg/ft ²) | 84.7 | 92.0 | 90.5 | 85.5 | | | |

(Continued next page)

Table 16 (Cont.). Solar Efficiency versus Silver Thickness

| Type Glass and Thickness in millimeters (inches) | Specimen No. | Silver Deposition in grams per square meter | Beckman Spectro Photometer (Over Solar Spectrum) | Specular Photometer (550 nm) | | | Specular Spectrophotometer (Over Solar Spectrum) | | |
|--|--------------|---|--|------------------------------|--------|--------|--|--------|--------|
| | | | 125 mrad | 16 mrad | 8 mrad | 4 mrad | 16 mrad | 8 mrad | 4 mrad |
| ASG Industries Float Glass 2.4 mm (3/32-inch) | 87.1 | 0.65 (60 mg/ft ²) | 85.0 | 92.0 | 92.0 | 91.0 | | | |
| | 88.1 | 0.86 (80 mg/ft ²) | 85.0 | 93.0 | 92.0 | 92.0 | | | |
| | 89.1 | 1.01 (94 mg/ft ²) | 86.0 | 93.0 | 92.0 | 88.0 | 83.6 | 83.2 | 77.2 |
| | 90.1 | 1.08 (100 mg/ft ²) | 85.0 | 93.0 | 93.0 | 93.0 | | | |
| | 91.1 | 1.29 (120 mg/ft ²) | 85.0 | 92.0 | 92.0 | 92.0 | | | |

Table 17. Properties of Coated Chemically Deposited Silver Mirrors (Ref. 103)

| Specimen No. | Company | Coating | Measuring Instrument | Specularity | Efficiency at the following Wavelengths in Nanometers | | | | | | | | | | Avg Over Solar Spectrum |
|--------------|-----------------|-----------|-----------------------------|-------------|---|-----|-----|-----|-----|-----|-----|------|------|------|-------------------------|
| | | | | | 426 | 498 | 561 | 623 | 691 | 774 | 860 | 1008 | 1208 | 1594 | |
| 99-2 | DuPont | 500 IMRON | Specular Spectro Photometer | 16 | 60 | 79 | 85 | 86 | 90 | 90 | 91 | 93 | 93 | 83 | 85 |
| | | | | 8 | 58 | 77 | 83 | 85 | 86 | 89 | 89 | 91 | 91 | 82 | 83 |
| | | | | 4 | 55 | 73 | 79 | 83 | 83 | 85 | 86 | 88 | 88 | 81 | 80 |
| | | | Specular Photometer | 4 | | | 83 | | | | | | | | |
| 102-2 | Sierracin | FX-103 | Specular Spectro Photometer | 16 | 60 | 80 | 88 | 90 | 93 | 94 | 95 | 98 | 97 | 75 | 87 |
| | | | | 8 | 59 | 80 | 87 | 90 | 92 | 93 | 95 | 96 | 95 | 75 | 86 |
| | | | | 4 | 59 | 80 | 87 | 89 | 90 | 92 | 95 | 96 | 94 | 71 | 85 |
| | | | Specular Photometer | 4 | | | 82 | | | | | | | | |
| 104-1 | Textar Plastics | C-254 | Specular Spectro Photometer | 16 | 58 | 76 | 84 | 90 | 90 | 92 | 93 | 96 | 96 | 87 | 86 |
| | | | | 8 | 58 | 75 | 83 | 88 | 89 | 92 | 93 | 95 | 96 | 87 | 86 |
| | | | | 4 | 57 | 75 | 82 | 87 | 88 | 90 | 93 | 94 | 96 | 80 | 84 |
| | | | Specular Photometer | 4 | | | 82 | | | | | | | | |
| 101-2 | DuPont | RK3637 | Specular Spectro Photometer | 16 | 57 | 79 | 86 | 90 | 90 | 92 | 93 | 95 | 96 | 90 | 87 |
| | | | | 8 | 56 | 77 | 85 | 88 | 90 | 91 | 91 | 94 | 94 | 88 | 85 |
| | | | | 4 | 56 | 77 | 83 | 88 | 90 | 90 | 90 | 92 | 94 | 83 | 84 |
| | | | Specular Photometer | 4 | | | 84 | | | | | | | | |

Table 18. Reflectance Degradation of Desert Mirrors Exposed for 184 Days (Ref. 103)

| Supplier | Coating Identification | Initial Values at Various Acceptance Angles in Milliradians (mrad) | | | | Exposure at Fort Irwin in Mojave Desert | |
|-------------------|------------------------|--|--|---------|--------|---|--------------------|
| | | Over Solar Spectrum using Beckman | At Wavelength of 550 nanometer using Specular Photometer | | | | |
| | | | 125 mrad | 16 mrad | 8 mrad | 4 mrad | Exposure (days) |
| ROHM and HAAS Co. | 70-2 (R&H No. 1) | 93.5 | 87.0 | 83.5 | 75.0 | 165 | Coating Failed |
| | 71-2 (R&H No. 2) | 96.8 | 95.5 | 95.0 | 94.0 | 166 | Coating Failed |
| | 72-2 (R&H No. 3) | 97.3 | 96.5 | 95.5 | 92.6 | 184 | Corrosion Starting |
| | 73-1 (R&H No. 4) | 97.2 | 96.5 | 95.0 | 92.0 | 184 | Corrosion Starting |
| | 74-2 (R&H No. 5) | 97.6 | 94.0 | 90.5 | 82.0 | 184 | Slight Crazing |
| | 76-1 (R&H No. 6) | 97.1 | 96.0 | 96.5 | 94.0 | 184 | Extensive Crazing |
| | 76-1 (R&H No. 7) | 97.1 | 95.0 | 94.5 | 92.5 | 184 | Slight Crazing |
| | 77-2 (R&H No. 8) | 96.8 | 93.0 | 90.5 | 82.5 | 184 | Extensive Crazing |
| | 78-1 (R&H No. 9) | 97.6 | 95.0 | 94.5 | 92.0 | 184 | Extensive Crazing |
| | 79-2 (R&H No. 10) | 97.9 | 94.0 | 92.5 | 84.5 | 185 | Coating Failed |
| | 80-1 (R&H No. 11) | 97.6 | 95.5 | 95.0 | 87.0 | 184 | Severe Corrosion |
| | 81-1 (R&H No. 12) | 95.4 | 90.0 | 89.0 | 85.0 | 184 | Moderate Crazing |
| O'Brien Corp. | 69-1 (6-73C) | 95.4 | 93.0 | 90.5 | 82.5 | 165 | Coating Failed |
| | 68-1 (874-C-200) | 96.3 | 90.5 | 87.0 | 71.0 | 184 | Coating Failed |

Table 19. Reflectance Degradation as a Function of Wavelength for Desert Soiled Mirrors (Ref. 103)

| Specimen Number | Mirror Glass | Mirror Condition | Instrument | Specularity | Solar Reflectance Efficiency at the Following Wavelengths in Nanometers | | | | | | | | | | Average Value Over Solar Spectrum | |
|-----------------|-------------------------------------|------------------|-----------------------------|-------------|---|------|------|------|------|------|------|------|------|------|-----------------------------------|--|
| | | | | | 426 | 498 | 561 | 623 | 691 | 774 | 860 | 1008 | 1208 | 1594 | | |
| 65.16 | ASG Ind Lustra Sheet 3.2 mm (0.125) | Dirty | Beckman | 125 | 92.5 | 95.5 | 96.5 | 96.0 | 94.5 | 92.0 | 90.5 | 89.5 | 89.5 | 92.0 | 92.9 | |
| | | | Specular Spectro Photometer | 16 | 83.6 | 87.7 | 89.4 | 88.5 | 92.2 | 86.8 | 85.4 | 83.9 | 84.7 | 89.9 | 87.0 | |
| | | | | 8 | 83.6 | 87.7 | 89.4 | 88.5 | 90.1 | 86.8 | 85.4 | 83.9 | 84.7 | 89.2 | 87.0 | |
| | | | | 4 | 79.6 | 81.7 | 81.7 | 82.1 | 82.8 | 82.4 | 81.2 | 79.2 | 77.9 | 73.0 | 80.0 | |
| | | | Specular Photometer | 16 | | | 96.5 | | | | | | | | | |
| | | | | 8 | | | 96.0 | | | | | | | | | |
| | | 4 | | | | 95.5 | | | | | | | | | | |
| | | Clean | Beckman | 125 | 94.0 | 96.5 | 98.0 | 97.0 | 96.5 | 94.5 | 92.0 | 90.5 | 90.5 | 93.5 | 94.3 | |
| | | | Specular Spectro Photometer | 16 | 90.9 | 93.7 | 94.5 | 93.7 | 95.2 | 91.7 | 90.5 | 89.8 | 89.5 | 94.6 | 92.0 | |
| | | | | 8 | 91.1 | 93.7 | 95.6 | 94.4 | 92.5 | 90.0 | 89.4 | 80.2 | 89.5 | 93.8 | 93.0 | |
| | | | | 4 | 89.1 | 93.1 | 94.0 | 93.8 | 94.3 | 80.0 | 83.8 | 82.6 | 80.5 | 79.6 | 87.0 | |
| | | | Specular Photometer | 16 | | | 95.0 | | | | | | | | | |
| 8 | | | | | 95.0 | | | | | | | | | | | |
| 4 | | | | 94.0 | | | | | | | | | | | | |

Table 20. Washing of Model Heliostat Using Different Washing Solutions
(Ref. 103)

| Company | Exposure Time (days) | Reflectance Efficiency (%) | | Washing Solution Application | | | | | Rinse Solution Application | | | | Drying Time (min) |
|---------|----------------------|----------------------------|-------|------------------------------|--------|-----------------|-----------|------------------|----------------------------|------------|-----------------|-----------|-------------------|
| | | Dirty | Clean | Pressure (psi) | Time | Quantity (gal.) | Nozzle | Dwell Time (sec) | Pressure (psi) | Time (sec) | Quantity (gal.) | Nozzle | |
| | | | | | | | | | | | | | |
| McGean | 20 | 59.1 | 64.75 | 80 | 35 sec | 0.83 | 80.10 | 30 | 80 | 90 | 3.2 | 65.16 | 20 |
| Turco | 29 | 56.52 | 63.54 | 40 | 3 min | 2 | 80.04 | 60 | 50 | 105 | 4.0 | 90.20 | 20 |
| TEC | 39 | N/A | N/A | 150 | 80 sec | 4 | Graco Gun | 60 | 150 | 145 | 5.0 | Graco Gun | 30 |

Table 21. Washing of SRE Heliostats using McGean Chemical's Washing Solution(Ref. 103)

| Heliostat No.* | Prewash Reflectance Efficiency (percent) | Postwash Reflectance Efficiency (percent) | Application Time | | Solution Quantity | | Solution Type | | Nozzle Size | |
|----------------|--|---|------------------|-------------|-------------------|--------------|---------------|-----------------|-------------|-------------|
| | | | Wash (min) | Rinse (min) | Wash (gal.) | Rinse (gal.) | Wash | Rinse | Wash (gpm) | Rinse (gpm) |
| | | | | | | | | | | |
| H ₁ | 65.9 | 78.2 | 1.0 | 5.0 | 1.50 | 14.0 | A69M | Deionized Water | 1 | 5 |
| H ₂ | 56.1 | 76.5 | 1.0 | 3.7 | 1.25 | 8.75 | A69M | Deionized Water | 1 | 5 |
| H ₃ | 69.2 | 87.5 | 1.0 | 3.0 | 0.75 | 8.0 | A69M | Deionized Water | 1 | 5 |
| H ₄ | 73.3 | 84.0 | 1.0 | 2.0 | 1.25 | 5.75 | CB120 | Deionized Water | 1 | 5 |
| IH1 | $\frac{76.8}{72.2}$ | $\frac{85.1}{86.6}$ | 1.4 | 2.8 | 1.60 | 7.75 | CB120 | Deionized Water | 1 | 5 |

* H₁, H₂, H₃ = Acrylic first surface mirrors, H₄ = Laminated mirror, IH1 = 3/32 mirror bonded to foam core laminated mirror

Table 22. Summary Table -- Identification of Reflective Surfaces for Solar Concentrator Applications

| Type of Reflective Surface | Thermal Expansion | Reflectance | Durability | Price | Present Availability |
|--|-------------------|--------------------|--------------------|----------------|----------------------|
| Glass, fusion Type 0317 | Superior | Superior | Excellent | Good | Superior |
| Glass, fusion Type 7806 (Modified) | NA [†] | Superior (Est.) | Superior (Est.) | Good (Est.) | NA |
| Glass, Schott Type B270 | Superior | Superior | Excellent | Superior | Good |
| Glass, float [‡] | Good | Good | Superior | Good | Excellent |
| Aluminum | Fair* | Superior** Good | Good | Fair | Superior |

* For cellular glass substrate

** If overcoated

† NA = Not available

‡ Recent preliminary measurements indicates that the characteristics of float glass may be approaching those of fusion glass.

SECTION VIII

REFERENCES

1. Hammond, A. L., "Energy and the Future," AAAS, Washington, DC, 1973, p. 61.
2. Noll, E. M., "Wind, Solar Energy," Howard W. Sams & Co., Inc, 1975.
3. Meinel, A. B. and Meinel, M. P., "Applied Solar Energy," Addison-Wesley Publishing Co., Reading, MA, 1977.
4. Duffie, J. A., and Beckman, W. A., "Solar Energy Solar Processes."
5. Considine, D. M., "Energy Technology Handbook," McGraw-Hill Book Co., Inc., 1977.
6. McGraw-Hill Encyclopedia of Science and Technology, McGraw-Hill Book Co., Inc., Vol. 11, 1977, p. 420.
7. "Properties of Glasses and Glass-Ceramics," Corning Glass Works, Corning, NY, August 1973.
8. Strand, E. B., "Glass Engineering Handbook," 2nd Ed., McGraw-Hill Book Co., Inc., NY, 1958.
9. Mantell, C. L., "Engineering Materials Handbook," McGraw-Hill Book Co., Inc., 1958.
10. Baumeister, T., Ed., "Mechanical Engineer's Handbook," McGraw-Hill Book Co., Inc., 1958, pp. 6-182.
11. Weiss, G., "The Book of Glass," Praeger Publ., Pall Mall Press Ltd., London, 1971.
12. Knittle, R. M., "Early American Glass," Garden City Publishing Co., Inc., NY, 1927.
13. Lee, R. W., "Early American Pressed Glass," R. W. Publishers, Northboro, MA, 1946.
14. McKearin, H., and McKearin, G. S., "American Glass," Crown Publ., NY, 1963.
15. Moore, N. H. "Old Glass, European and American," Tudor Publishing Co., NY, 1935.
16. Clauser, H. R., "The Encyclopedia of Engineering Materials and Processes," Reinhold Publishing Corp., Chapman & Hall, NY, 1963.
17. Bolz, R. E., and Tuve, G. L., "Handbook of Tables for Applied Engineering Science," Chemical Rubber Co., Cleveland, Ohio, 1970.

18. Weast, R. C. and Astle, M. J., "CRC Handbook of Chemistry and Physics," 59th Ed. CRC Press Inc., West Palm Beach, FL, 1978.
19. Morey, G. W., "Properties of Glass," Rheinhold Publishing Co., 1938.
20. Hutchins, J. R. and Harrington, R. V., "Glass", Encyclopedia of Chemical Technology, 2nd Ed., Vol. 10, John Wiley & Sons Inc., 1966.
21. Scholes, S. R., "Modern Glass Practice," International Publications Inc., Chicago, IL, 1952.
22. Harper, C. A., "Handbook of Electronic Packaging," Vol. 7, McGraw-Hill Book Co., Inc., NY, 1965, p. 114.
23. Horton, F. H., "Elements of Ceramics," Addison-Wesley Press, Inc., Cambridge, MA, 1952.
24. ASTM Standards of Glass and Glass Products, April 1955.
25. Dube, G., "Glass for Fusion Research," Optical Spectra, May 1978, pp. 40-45.
26. Dart, William C., "Glass for Industry," Optical Spectra, October 1977, pp. 46-49.
27. Solar Glass Mirror Program, Report of the Solar Glass Mirror Development Coordinating Committee, October 16, 1978.
28. Van Vlack, L. H., "Elements of Material Science," Addison-Wesley Publishing Inc., Reading, MA, 1964.
29. Williams, C. S. and Becklund, O. A., "Optics," Wiley-Interscience, NY, 1972.
30. Stadler, L. E., and DaDue, A. W., "Review Shows Status of Glass Compositions," Glass Industry, December, 1972, pp. 14-16, and January, 1973, pp. 13-15.
31. Tooley, F. V., Editor, "The Handbook of Glass Manufacture," Vol. II, Books for Industry, Inc., and Glass Industry Magazine, Divisions of Magazines for Industry, Inc., 1974, p. 744.
32. Gaiser, R. A., Lyon, K. C., and Scholes, A. B., "Glass Surface Protection System Sprays Hot and Cold," Ceramic Industry, 84 (4), 1965, pp. 96-100, 136-40.
33. Evans, C., "Strengthening Glass With Tin Chemicals," Glass, 51 (9), 1974 pp. 303-5.
34. Materials Issue, Scientific American, Vol. 216, No. 9, September, 1967.
35. Greene, C. H., Scientific American, Vol. 204, No. 1, January, 1961, p. 92.
36. Lillie, H. R., "Glass," Chapter 7, Handbook of Physics, McGraw-Hill Book Co., Inc., NY, 1958.

37. Slater, J. C. "Quantum Theory of Matter," McGraw-Hill Book Co., Inc., 1951, p. 304.
38. Younger, P. R., "Integral Glass Encapsulation for Solar Arrays," JPL 5101-94, 11th Project Integration Meeting, December 13-14, 1978.
39. Flinn, R. A. and Trojan, P. K., "Engineering Materials and their Applications," Houghton Mifflin Co., Boston, 1975.
40. Hench, L. L., Dove, D. B., and Loehman, R. E., "Evaluations of Laser Window Materials", Florida Univ., Gainesville Engineering and Industrial Experiment Station, Final Report 1 July 1971 - 30 June 1972, Contract F19628-71-C-0256, DDC Report No. AD-906 561, August, 1972.
41. Pliskin, W. A., Kerr, D. R., and Perri, J. A., "Thin Glass Films," in Physics of Thin Films, 4, Academic Press, 1967, pp. 257-324.
42. Wiser, G. L., "New Materials in Aircraft Windshields," SAE Paper 700862, Presented at National Aeronautic and Space Engineering and Manufacturing Meeting, October, 1970, Los Angeles, CA. p. 519.
43. Olson, J. B., "Design Considerations Affecting Performance of Glass/Plastic Windshields in Airline Service," Aircraft Engineering, Vol. 47, November, 1975, pp. 4-12.
44. Wiser, G. L., "Glass/Plastic Composite Windshields," SAMPE Journal, 6, 1972, pp. 29-35.
45. Jones, G. O., "Glass," Chapman and Hall Ltd., 1971.
46. Southall, J. P., "Mirrors, Prisms, Lenses," 1964.
47. Wilson, Mitchell, "Energy," Time Incorp., NY, 1963, p. 90.
48. Jurgen R. Meyer-Arendt, "Introduction to Classical and Modern Optics," Prentice-Hall, Inc., 1972.
49. Marcuse, D., Editor, Integrated Optics, IEEE Press, 1972.
50. Harris, J. H., et al., Optical Waveguide Scattering and Griffith Microcracks, Integrated Optics, IEEE Press, pp. 181-182.
51. Stratton, J. A., "Electromagnetic Theory," New York, McGraw-Hill, 1954.
52. Ernsberger, F. M., "A Study of the Origin and Frequency of Occurrence of Griffith Microcracks on Glass Surfaces," Research Into Glass, PPG Industries Science Press, 1967.
53. Gray, D. E., Editor, "American Institute of Physics Handbook," McGraw-Hill Book Co., Inc., NY, 1963.
54. Born, M. and Wolf, E., "Principles of Optics," Pergamon Press, Inc., New York, 1959.

55. Johnson, C. C., "Field and Wave Electrodynamics," McGraw-Hill Book Co., Inc., NY, 1965.
56. Goodman, J. W., "Introduction to Fourier Optics," McGraw-Hill Book Co., Inc., NY, 1968.
57. Data Sheets, Liberty Mirror Co., Brackenridge, PA, September, 1978.
58. ASTM Index, Vol. P 48., Philadelphia, PA, 1976.
59. Eighth International Congress on Glass, Society of Glass Technology, England, 1969.
60. Walker, B. H., The 20 Percent Rule: Selecting Optical Glass., Optical Spectra, Vol. 12 (12) December 1978, pp 42-43.
61. Phillips, C. J., "Glass the Miracle Maker," 2nd Edit., Pitman Publishing Corp., NY, 1948.
62. Pincus, A. G., "Invitation to Glass Technology," Ceramic Age, Vol. 53, 5, 260; Vol. 54, 1, 19; Vol. 54, 2, 81, 1949.
63. Weyl, W. A., "Glass Composition," Ceramic Age, Vol. 50, 1947, p. 54
64. Davis, P., "The Development of the American Glass Industry," Harvard University Press, Cambridge, MA, 1949.
65. Devillers, R. W., and Vaerewyck, F. E., "Glass Tank Furnaces," trans. by Scholes. Ogden-Whitney Press, NY, 1937.
66. "Glass Glossary." Bull. Am. Ceram. Soc. Vol. 27, 1948, p. 355.
67. Howard, G. E., "Mechanization of the Glass Industry." Glass Industry, I, (31), 75, 1950; II, (31), 131, 1950; III, (31), 183, 1950.
68. Boenig, H. V., "Polyolefins, Structure and Properties," Elsevier Publ. Co., NY, 1966.
69. Muller, K. and Rabu, E., "Fundamentals of Metal Deposition," Ibid, 1967.
70. Rodgers, T. M., "Handbook of Practical Electroplating," The MacMillan Co., NY, 1959.
71. Campbell, F. J., "Effects of Radiation on Transmittance of Glasses and Adhesives," Proceedings of 17th Annual Power Sources Conference, May 21-23, 1963.
72. Illes, P. A., "Integral Glass Coatings for Solar Cells," NASA-CR-57963, 1964.

73. Wallis, G., and Pomerantz, D. I., "Field Assisted Glass-Metal Sealing," J. Appl. Phys., 40 (10), 1969, pp. 3946-49.
74. Wallis, G., "Glass-to-Metal Seals Without Adhesives," Automotive Engineering, 79 (2), 1971, pp. 15-19.
75. "Solar Cell Cover Glass Development," Ion Physics Corp., NASA-CR-97123, October 1, 1967.
76. Ward, J., "Towards Invisible Glass," Vacuum, 22 (9), September, 1972, pp. 369-75.
77. Holland, L., Vacuum Deposition of Thin Films, Chapman & Hall, Ltd., 1963.
78. Luft, W., "Status of TiO_x Antireflective Coating in U.S.," Conference Record of the Tenth IEEE Photovoltaic Specialists Conference, Palo Alto, CA, November 13-15, 1973.
79. Nicoll, F. H., "A New Chemical Method of Reducing the Reflectance of Glass," RCA Review, 6, 1942, pp. 287-301.
80. Hollahan, J. R., and Wydeven, T., "Protection of Moisture Sensitive Optical Components With Polymer Coating," U.S. Patent Application 402,866, February 10, 1973.
81. Shirokshina, Z. V., "Antireflection and Protective Coatings of Components Made From Lead Germanate Glasses," Sov. J. Opt. Technol. (USA), 41 (9), September, 1974, pp. 411-12.
82. Stachiw, J. D., "Acrylic Plastics in Hydrospace," Oceanol. Int., 5 (2), February, 1970, pp. 24-27.
83. Haynos, J. G., "Investigation of Resinous Materials for Use as Solar Cell Cover Glass Adhesive," NASA-TM-X-55333, X-716-65-369, September 1965.
84. Carmichael, D. C. et al., Review of World Experience and Properties of Materials for Encapsulation of Terrestrial Photovoltaic Arrays, Final Report, ERDA/JPL-954328-76/4 Battelle (Columbus) Laboratories, July 21, 1976.
85. Cottardi, V., "Analysis of Thin Films on Glass by Nuclear Techniques," Glass Technology, Vol. 17, No. 1, February 1976, pp. 26-34.
86. "The Chemical Durability of Glass," Glass Technology, Vol. 17, No. 6, December 1976, pp. 131-135.
87. Ezeilo, C. C. O., "The Effect of Dropwise Condensation on Glass Solar Properties," Solar Energy, Vol. 19, No. 4, 1977, pp. 389-394.
88. Weyl, W. A., "Coloured Glasses," Society of Glass Technology, Sheffield, S105B5, England, 1976.

89. "Studies of Encapsulation Materials for Terrestrial Photovoltaic Arrays," JPL Contract 954328, Battelle, Columbus Laboratories, First Quarterly Progress Report (December 22, 1975); Second Quarterly Progress Report, ERDA/JPL-954328-76/1 (March 26, 1976); Third Quarterly Progress Report, ERDA/JPL-954328-76/2 (June 28, 1976).
90. Yasui, R. K., and Goldsmith, J. V., "Status of JPL Solar Powered Experiments for Terrestrial Solar Power Applications," Presented at Ninth Intersociety Energy Conversion Engineering Conference, San Francisco, August 26-30, 1974.
91. Yasui, R. K., and Patterson, R. E., "Utilization of Space Technology for Terrestrial Solar Power Applications," Conference Record of the Tenth IEEE Photovoltaic Specialists Conference, November 13, 15, 1973, Palo Alto, California, 1976.
92. Besson, J., Nguyen Duy, T., Gauthier, A., Palz, W., Martin, C., and Vedel, J., "Evaluation of CdS Solar Cells as Future Contender for Large Scale Electricity Production," The Conference Record of the Eleventh IEEE Photovoltaic Specialists Conference, 1975, May 6-8, 1975, Scottsdale, Arizona, 1975.
93. Pullmanov, N. V., and Potapov, V. N., "Solar Batteries in Protective Transparent Covers," *Geliotekhnika*, 9 (5), 1972 (UDC 662.997.62) pp.25-28.
94. Thermal Power Systems, Adv. Subsystem Development, 2nd Semi-Annual Progress Report, DOE/JPL-1060-7816, November, 1978.
95. Freese, J. M., The Development of a Portable Specular Reflectometer for Field Measurements of Solar Mirror Materials, November, 1978.
96. Lind, M. A., Heliostat Glass Survey and Evaluation, DOE Semiannual Review, Dallas, Texas, September 19-21, 1978.
97. Grimmer, D. P., et al., Augmented Solar Energy Collection Using Various Planar Reflective Surfaces: UC-62, LA 7041, April, 1978.
98. Butler, B. L., and Pettit, R. B., "Optical Evaluation Techniques for Reflecting Solar Concentrators," *SPIE*, Vol. 114, 1977.
99. Pettit, R. B., Characterization of the Reflected Beam Profile of Solar Mirror Materials, *Solar Energy*, Vol. 19, 1977, pp. 733-41.
100. Banas, J. F., Technical Status Report, Solar Energy Project Management, July 1978.
101. Wilson, W. G., "Heliostat Technology Studies," Sandia Lab., Livermore, CA, September 21, 1978.
102. "Solar Collector Reflective Surface," G.E., STE-LSE Shenandoah Design Review, August 18, 1978.
103. Taketani, H., et al., "Mirrors for Solar Energy Application," McDonnell Douglas Astro. Co., West, MDC G 7213, September 1977.

104. Jenkins, F. A., and White, E., "Fundamentals of Optics," 4th Edition, McGraw-Hill Book Co., Inc., NY, 1976.
105. JPL Document on Soiling of Reflective Surfaces (in preparation).
106. Kazanchyan, G. P., "Weathering Properties of Polymer Materials Exposed to Intense Solar Radiation," *Geliotekhnika*, Vol. 1, No. 5, 1965, pp. 40-44.
107. Rittenhouse, J. B., and Singletary, J. B., "Space Materials Handbook," Supplement 1 to 2nd Edition, "Space Materials Experience," 1966.
108. Lillywhite, M. A., "Low-Energy Proton Effects on Solar Cell Cover Glass Materials," Vol. 1, NASA-CR-73372, November, 1969.
109. Lillywhite, M. A., "Low-Energy Proton Effects on Solar Cell Cover Glass Materials," Vol. 2, NASA-CR-73438, November, 1969.
110. Doremus, R. H., "Glass Science," John Wiley & Sons, Inc., NY, 1973, pp.115.
111. Data from R. B. Pettit, Sandia Corp. Laboratories, Albuquerque, NM.
112. Wen, L. C., Thermal Optical Surface Properties and High Temperature Solar Energy Conversion, 2nd AIAA/ASME Thermophysics and Heat Transfer Conference, Palo Alto, CA, May 24-26, 1978.
113. Walters, H. V. and Adams, P. B., "Effects of Humidity on the Weathering of Glass," *Journal of Non-Crystalline Solids*, 19, 1975, pp. 183-99.
114. Salama, M. A. and Bouquet, F. L., On the Thermoelastic Analysis of Solar Cell Arrays and Related Materials, TM 33-753, JPL, California Institute of Technology, February 15, 1976.
115. Communication with J. M. Freeze, Sandia Corp., Albuquerque, NM.
116. Wray, J. H., and Neu, J. T., *J. Optical Soc. Am.* 59, 1969, pp.774.
117. Hass, G., in Kingslake, R., ed., "Applied Optics and Optical Engineering," Vol. III, Academic Press, Inc., NY, 1965, p. 309.
118. Hodgman, C. D., "Handbook of Chemistry and Physics," Chemical Rubber Co., 13th Edition, pp.1751, 1947.
119. Hockok, F., "Handbook of Solar and Wind Energy," Cahner Books, 1975.
120. Stanworth, J. E., "Physical Properties of Glass," Oxford University Press, Amen House, London, E.C.4, 1950.
121. Bouquet, F. L., "Aging Characteristics of Mirrors for Solar Energy Applications," DOE/JPL 1060-19 (JPL Internal Document 5102-116), April, 1979.